Week 5

Optical Properties 3 Exciton Diffusion Chapter 3.9-3.10



Exciton Diffusion

- The macroscopic transfer of energy resulting from a series of exciton transfers between molecules
 - Process is random
 - Duration of process determined by lifetime $\tau_{\rm D}$ of the exciton
 - Mean distance travelled known as diffusion length, L_D
 - Diffusion constant:

$$L_D^2 = D\tau_D$$



Capture of excitons by acceptors

Diffusion over large distances occurs by a series of random transfer steps from donor to acceptor

→ Relationship between capture rate, k_c , and the diffusion constant is obtained using the diffusion equation

→Assume probability for capture is unity when the exciton arrives within a distance

 R_c from the acceptor. (R_c = exciton *capture radius*.)

$$k_c \approx 4\pi DR_c$$

Chandrasekhar, S. 1943. Rev. Modern Phys., 15, 1.

If diffusion occurs by individual FRET steps, then the relationship between L_D and the Förster radius is:

$$L_{D,FRET} = \frac{1}{\sqrt{6}} \frac{R_0^3}{a^2}$$

- Assumes molecules on a 3D cubic lattice, lattice constant *a*.
- Analogous relationship holds for Dexter transfer

Diffusion Equation



1

Measured Diffusion Lengths

Lunt, R. R., et al. J. Appl. Phys., 105, 053711.

Table 3.6: Measured diffusion lengths (L_D) for singlet (S) and triplet (T) excitons of crystalline (C.) and amorphous (A.) films measured by spectrally resolved photoluminescence quenching.

Material ^{a,b}	Exciton	Morphology ^c	Quenching/Blocking Layer ^b	$L_D(\mathrm{nm})$	D (10 ⁻⁴ cm ² /s)
NPD	S	А.	C ₆₀ /BCP	5.1 (±1.0)	0.7 (±0.2)
CBP	S	А.	C ₆₀ (or NTCDA)/ Free	16.8 (±0.8)	40 (±12)
SubPc	S	А.	C ₆₀ /Bare	8.0 (±0.3)	≥6.4
PTCDA	S	C 55nm (flat)	C ₆₀ (or NPD)/ NTCDA	10.4 (±1.0)	3.4 (±0.9)
DIP	S	C >150nm (up)	C ₆₀ / Free	16.5 (±0.4)	15 (±4)
DIP	S	C 30nm (flat)	C ₆₀ / Free	21.8 (±0.6)	26 (±7)
C ₆₀	S	А.	NPD/BPhen	34 (±3)	20 (±4)
C ₇₀	S	А.	NPD/BPhen	10 (±2)	
PtOEP	T - Mon.	C >150nm (up)	C ₆₀ /BCP	18.0 (±0.6)	0.041
					(±0.003)
PtOEP	T – Dim.	C >150nm (up)	C ₆₀ /BCP	13.1 (±0.5)	0.00061
					(±0.0001)

^a All data are from (Lunt et al., 2009) except for C_{60} and C_{70} from (Bergemann et al., 2015). The larger error bars for L_D for C_{60} and C_{70} is due to their comparatively weak luminescence.

^b Materials used: NPD= N,N'-diphenyl-N,N'-bis(1-naphthyl)-1,1'biphenyl-4,4''diamine; CBP=4'-bis(9-carbazolyl)-2,2'-biphenyl; PTCDA=3,4,9,10-perylenetetracarboxylic dianhydride; NTCDA=3,4,7,8 naphthalenetetracarboxylic dianhydride; SubPc= boron subphthalyocyanine chloride; DIP= diindenoperylene; PtOEP=Pt octaethylporphorin; BCP=bathocuproine; BPhen=bathophenanthroline; Free=no layer.

^c Up/flat refers to whether the molecular plane is perpendicular/parallel to the substrate.



Annihilation!

When excitons collide at a single molecular site, they can combine to form a single exciton and a ground state molecule: <u>Exciton-exciton annihilation</u>

Spin multiplicity can determine annihilation efficiency (Dexter vs FRET). Annihilation leads to losses, and phenomena like <u>delayed fluorescence</u>.

When excitons collide with charges, the charge can be excited, losing the exciton: <u>Exciton-polaron annihilation</u>





Bad things happen to good excitons

 $S+T \xrightarrow{k_{STA}} T^n + S_0 \xrightarrow{k_{Tn}} T + S_0$ (STA) $\begin{array}{c} k_{SPA} & k_{Pn} \\ S + P \xrightarrow{} P^{n} + S_{0} \xrightarrow{} P + S_{0} \end{array}$ (SPA) S_1/T_1 S_1/T_1 energy transfer $\begin{array}{c} k_{SSA} & k_{Sn} \\ S+S \xrightarrow{} S^n + S_n \xrightarrow{} S + S_n \end{array}$ (SSA) SSA, TTA, STA 0 Delayed fluorescence S_1/T_1 $T + T \xrightarrow{k_{TTA-S}} S^n + S_0 \xrightarrow{k_{Sn}} S + S_0 \quad \text{(TTA-S)}$ **Triplet fusion** energy transfer S $T + T \xrightarrow{k_{TTA-T}} T^n + S_0 \xrightarrow{k_{Tn}} T + S_0 \quad \text{(TTA-T)}$ SPA, TPA 2 $T + P \xrightarrow{k_{TPA}} P^n + S_0 \xrightarrow{k_{Pn}} P + S_0$ (TPA) Organic Electronics Stephen R. Forrest Singlet fission when $S \rightarrow 2T$ $E_{s} \geq 2E_{T}$

What we learned

- The concepts leading to our understanding of individual molecular spectra
 - Born-Oppenheimer, Franck-Condon, Pauli Exclusion
 - Molecular orbitals
- Understanding molecular spectra
 - Excitons
 - Fermi's Golden Rule
 - Selection rules
 - > Spin: Fluorescence, phosphorescence, delayed fluorescence
- Building up to a solid
 - > Dimers, excimers, exciplexes
 - Polarization effects in solution and the solid state
- Exciton diffusion and annihilation

The next step: Charge transport in solids

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Electronic Properties of Organic Semiconductors

Electronic Properties 1

Energy Bands Electron Transport and Mobility

Chapter 4.1, 4.2, 4.3.1, 4.3.2.1, 4.3.2.2



Objectives: Electronic Properties

"When talking about semiconductors, if you can't draw a band diagram then you don't know what you're talking about", 'Kroemer's Lemma', Herbert Kroemer, ca. 1990.

- Organic electronic devices can only be understood in the context of the conductive properties of materials
- In this discussion we introduce
 - Origins of electronic band structure
 - Concept of polarons (large and small)
 - Charge transfer
 - Conductivity, effective mass and mobility
 - Injection
 - Charge trapping
- At the end of this discussion, you will have learned about structure, optical and electronic properties: all the tools needed to fully understand and analyze all OE devices and phenomena



Modes of Conduction





- Coherent
- Charge mean free path $\lambda >> a$
- $BW > k_B T$, $\hbar \omega_0$

Hopping and tunneling transport



- Incoherent (each step independent of previous)
- Charge mean free path $\lambda \sim a$
- Tunneling between states of equal energy is band-like

•
$$BW < k_B T$$
, $\hbar \omega_0$

Hopping

Time

When there is strong electron-phonon (small polaron) coupling, we get another condition for band transport:

$$\mu > \frac{qa^2}{\hbar} \underbrace{\begin{pmatrix} \hbar \omega_0 \\ k_B T \end{pmatrix}}_{(typ. ~ 100 meV)}$$

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12

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 μ ~ 20 cm²/V-s at room temperature...very high (and probably unrealistic)!

Transport Bands in Organics

- **Tight binding** approximation is useful due to importance of only nearest neighbor interactions
- Recall case of dimers and larger aggregates on exciton spectrum. Close proximity of neighbors results in:
 - Coulomb repulsion
 - Pauli exclusion
 - Splitting leads to broadening of discrete energies into bands



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13

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Can we count on such a simple theory?

- Weak binding makes this approach plausible
- Nearest neighbor interactions (e.g. due to vdW forces) makes the solutions tractable.
- What does it predict?



Jayaraman, et al. 1985. J. Chem. Phys., 82, 1682.

Molecules under hydrostatic pressure have a decreasing energy gap.

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Measuring Band Structure (and other energies of interest)

- Ultraviolet photoelectron spectroscopy (UPS)
- Photoelectric effect used to measure energy of a single electron from the HOMO to the vacuum level.
- Gives *k* of photoemitted electron
- Varying angle (ARUPS) of sample gives dispersion E(**k**) for k_{\perp}





Interpretation of PES Spectra



SOMO=singly occupied MO



Calculated and Measured Band Structures



Calculated structure for biphenyl:

- Two molecules/cell give 2 branches along each direction
- Max. BW= ~ 70 meV for electrons & holes

Katz, J. I., et al. 1963. J. Chem. Phys., 39, 1683.



Hasegawa, S., et al. 1994. J. Chem. Phys., 100, 6969.

Measured structure for BTQBT:

- Technique: ARUPS
- Minimum contact distance: 3.26 Å
- HOMO BW = 400 meV
- m* = 3.1 m₀
- Recall: $\mu_h = \frac{q\tau}{m_h^*}$
- But thermally broadened bands have $\tau > \hbar / k_B T$
- $\mu \sim 6.5 \text{ cm}^2/\text{V-s}$ (c.f. Hall measurement of 4 cm²/V-s)

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17

(NHOMO = next highest orbital)

Charge Mobility Describes Transport in Solids

- Charge mobility: *μ* (not the dipole moment!)
- Definition: Constant of proportionality between velocity and electric field:

$$\mathbf{v}(\mathbf{k}) = \vec{\mu}_{\mathbf{k}} \mathbf{F}$$

- Tensor: dependent on crystal direction
- > Generally field dependent: $\mu = \mu(\mathbf{F})$
- > Depends on energy dispersion (i.e. band structure) via:

$$\mathbf{v}(\mathbf{k}) = \frac{1}{\hbar} \frac{\partial E(\mathbf{k})}{\partial \mathbf{k}}$$

For band-like transport: $\mu = \frac{q\tau}{m^*}$

(τ = mean free scattering time of the charge in the crystal: For thermally broadened bands: $\tau > \hbar / k_B T$)

• Ohms Law: $\mathbf{j} = q(n\mathbf{v}_e + p\mathbf{v}_h) = \ddot{\sigma}\mathbf{F}$

> or

$$\vec{\sigma} = q \left(n \vec{\mu}_e + p \vec{\mu}_p \right)$$



Mobility and Charge Diffusion

• <u>Near equilibrium</u>, the Einstein relationship connects these quantities:

$$\frac{D}{\mu} = \frac{k_B T}{q}$$

• Charge diffusion length: $L_q = \sqrt{D\tau}$

Band vs. Hopping Transport

- The charge diffusion length is: $L_q = \sqrt{D\tau} = \left[\frac{\mu\tau k_B T}{q}\right]^{\frac{1}{2}}$
- Band transport occurs when $L_q >> a$.
- From uncertainty: $BW \bullet \tau > \hbar$
- Condition for band conduction $\Rightarrow \mu > \frac{qa^2}{\hbar} \left(\frac{BW}{k_BT}\right)$
 - (Ex. Room temperature, let *BW*=25 meV, *a*=5Å. Then μ >5 cm²/V-s)
- Hopping due to short range interaction involving only nearest neighbor molecules
 - Incoherent diffusive process
 - Electron is heavy since it self traps: it polarizes the neighborhood and must carry that energy along with it.

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- Since only nearest neighbors are affected = small polaron
- Ionic materials, where the interaction goes as ~1/r = large polaron

Holstein, T. 1959. Studies of Polaron Motion. Part II. The "Small" Polaron. Ann. Phys., 8, 343?0

Hopping Formalism

• Total Hamiltonian:



• Important terms:

 $H_e^{tr} = \sum_{m \neq n}^N J_{nm} a_n^+ a_m$ $a^+(a) =$ electron creation (annihilation) operator

 $J_{nm} = \langle \psi_n(\mathbf{r} - \mathbf{R}_n) | H_e^{tr} | \psi_m(\mathbf{r} - \mathbf{R}_m) \rangle$ Overlap between molecules *m*, *n* - As in tight binding, leads to BW

 $\mathcal{N}_i = a_i^+ a_i$ Number operator

e-phonon coupling constant

$$H_{e-ph}^{loc} = \frac{1}{\sqrt{N}} \sum_{\mathbf{q},j} \sum_{m}^{N} \hbar \omega_{\mathbf{q},j} \Big(g_m(\mathbf{q},j) b_{\mathbf{q},j} + g_m^*(\mathbf{q},j) b_{-\mathbf{q},j}^+ \Big) a_m^* a_m$$

On-diagonal dynamic disorder: couples excess electron to molecular vibronic levels

And so on....

q = phonon wavevector *j* = phonon branch Organic Electronics Stephen R. Forrest

The case for static disorder

- Most organic semiconductors are permanently disordered
 - Polymers generally not formed into crystals
 - Small molecules used in devices are often amorphous or nanocrystalline
 - Even "perfect" crystals have impurities, stacking faults, dislocations
- A complete picture thus must include static disorder term:

$$H_e^{stat} = \sum_n^N \delta \varepsilon_n a_n^+ a_n + \sum_{\substack{m,n \ m \neq n}}^N \delta J_{nm} a_n^+ a_m$$

• Static disorder leads to hopping in both energy and space.



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Visualizing Lattice Distortions



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Small polaron theory predicts several transport regimes

- Dependent on coupling strength, g
- Static disorder not included here
- Band and tunneling both coherent and follow power law dependence



Going from formal theory to a practical quantity (mobility: µ)

- Diffusion constant is calculated from hops from site $p: p \rightarrow p \pm 1$
- Then in 3D: $D = \frac{1}{3}k_{ET}(p \rightarrow p \pm 1)\frac{a^2}{2}$ (6 sites to choose from on a cube)
- It follows from Einstein: $\mu = \frac{q}{3k_BT}k_{ET}(p \to p \pm 1)\frac{a^2}{2}$
- From small polaron theory we obtain k_{ET} (remember Fermi's Golden Rule!)
- And with Holstein's help, in the high temperature limit $(k_B T > \hbar \omega_0)$:

$$\mu_{hop} = \frac{qJ^2a^2}{6k_BT\hbar} \left[\frac{\pi}{2E_{pol}k_BT}\right]^{1/2} \exp\left(-\frac{E_{pol}}{2k_BT}\right)$$

- \checkmark The hopping mobility is thermally activated
- ✓ It scales with the square of the bandwidth, J^2 .
- ✓ As $E_{\rm pol}$ increases, $\mu_{\rm hop}$ decreases

✓ As
$$T \rightarrow$$
 large; then $\mu_{hop} \sim T^{-3/2}$

• As $T \rightarrow$ small; then μ_{hop} replaced by μ_{tun}



: scattering!

Molecular energy changes when a charge is transferred



Or polarization energy $\approx E_{pol}^{loc} = \lambda_{reorg} / 2$.



The theory of μ

- **Goal:** To find μ , but now with disorder expressed through $\rho(\varepsilon)$.
- We first need to find the rates, k_{ij} in the presence of disorder.
 - Model 1: Miller-Abrahams theory
 - Developed for impurity band conduction in semiconductors
 - Valid for weak electron-phonon coupling $(g_{m}, g_{nm} \text{ small})$

$$k_{ij} = v_0 \exp\left(-2\gamma_{ij}R_{ij}\right) \begin{cases} \exp\left(-\frac{\varepsilon_j - \varepsilon_i}{k_B T}\right) & \varepsilon_j > \varepsilon_i \\ 1 & \varepsilon_j < \varepsilon_i \end{cases} \quad \begin{cases} v_0 = hop_i \\ \gamma = overlap \\ R_{ij} = hop_i \end{cases}$$

 v₀= hopping attempt freq.~opt. phonon freq.
 γ=overlap factor, decay of wavefunction between *i*,*j*.
 R_{ii}=hopping distance

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- Implications:
 - "Uphill" transfers are thermally activated
 - "Downhill" transfers encounter no barrier
 - Valid when ε_j ε_i < Θ_m (Debye energy) of acoustic and optical phonons (~0.15 eV)
 ⇒ low temperatures
 - In F-field, add in $-q\mathbf{r} \cdot \mathbf{F}$ to exponential argument where F points from $j \Rightarrow i$

Miller, A. & Abrahams, E. 1960. Phys. Rev., 120, 745.

Model 2: Marcus Transfer

- Developed for understanding transfer of electron from donor to acceptor in solution
 - Generalized form of mobility in the small polaron (non disordered) model. Starting point will again be from Holstein:

$$k_{ET} = \frac{6k_B T \mu}{qa^2} \times \frac{J^2}{\hbar} \left[\frac{\pi}{2E_{pol}k_B T} \right]^{1/2} \exp\left(-\frac{E_{pol}}{2k_B T}\right)$$

- Valid for both upward and downward jumps: Only based on difference in free energy between initial and final states, ΔG .
- Valid at high temperatures, and strong electron-phonon couplings
- Activation energy for the transfer reaction: $2E_{act} = E_{pol} = \lambda_{reorg}/2$
- From Miller-Abrahams:

$$J = J_0 \exp\left(-\gamma_{ij} R_{ij}\right)$$

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Marcus, R. A. 1993. Rev. Modern Phys., 65, 599.

Transfer regimes under Marcus

Transfer rate depends on ΔG which can be less than or greater than 0.



Density of States and the Mobility Edge



$$\langle \varepsilon_{\infty} \rangle = \frac{\int_{-\infty}^{\infty} \varepsilon \rho(\varepsilon) \exp(-\varepsilon / k_B T) d\varepsilon}{\int_{-\infty}^{\infty} \rho(\varepsilon) \exp(-\varepsilon / k_B T) d\varepsilon} = -\frac{\sigma^2}{k_B T}$$

Mean equilibrium carrier energy— Charges relax into the tail of the distribution R. Forrest

Mobility (at last!)

- We now have everything we need:
 - Site occupation: $f_i(\varepsilon)$ from Fermi statistics
 - Transfer rate: k_{ij}
 - Site DOS: *ρ(ε)*
- As seen previously, mobility depends on k: $\mu = \frac{q}{3k_{BT}}k_{ET}(p \rightarrow p \pm 1)\frac{a^{2}}{2}$
- But this is not possible to solve exactly.
- Based on Marcus theory (most applicable), including electric field effects we get (ouch!):

$$\mu \approx \mu_0 \exp\left\{-\frac{E_{act}}{k_B T} - \frac{1}{8q^2} \left(\frac{\sigma}{k_B T}\right)^2 + \frac{1}{2\sqrt{2}q^2} \left[\left(\frac{\sigma}{k_B T}\right)^{3/2} - \left(\frac{\sigma}{k_B T}\right)^{1/2}\right] \sqrt{\frac{qaF}{\sigma}}\right\}$$

- At F=0, mobility activation ~ $1/T^2$ at low T, 1/T at high T
- Recall polaronic dependence follows ~1/T
- Monte-Carlo simulations show similar form

Fishchuk, I., et al. 2003. Phys. Rev., 67, 224303.

Poole-Frenkel type dependence Stephen R. Forrest