

Week 2-5

Light emitters 5

OLED Reliability

Lasers

Chapter 6.7-6.8



Organic Electronics
Stephen R. Forrest

Reliability Testing Methodologies

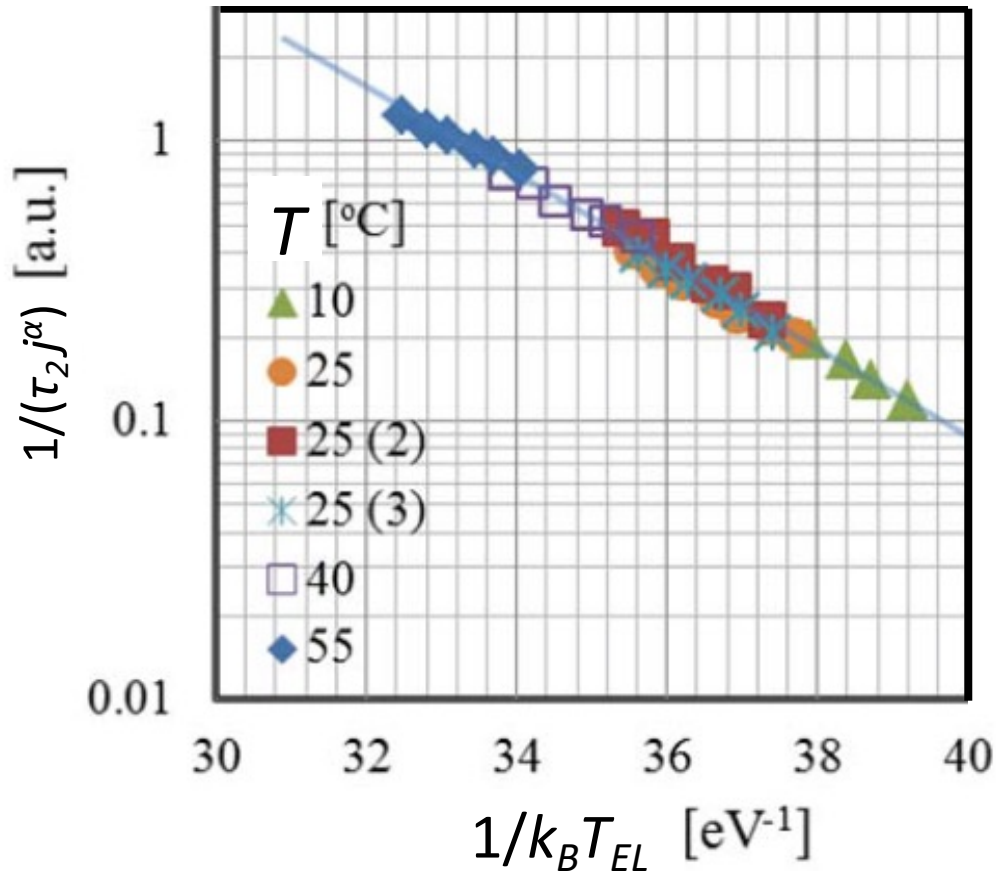
- Need to set clear metrics for failure
 - Example: Operating time for initial luminance (L_0) to decrease 10% from its initial value (called T90, or LT90)
 - Employ a population of equivalent devices and monitor their performance parameter (e.g. luminance) under normal operating conditions
 - If degradation slow, then an empirical degradation relationship is determined to extrapolate time to failure
 - Example: **Stretched exponential function**:
$$L(t) = L_0 \exp(-t/\tau)^\beta \quad \tau, \beta = \text{empirical constants}$$
- If degradation too slow, need to accelerate via increased T or L_0 .
 - Accelerated conditions must not introduce new failure modes
 - Need empirical relations to normalize lifetime to standard operating conditions (called **acceleration factors**)

$$LTx(L_0) = LTx(L_{0ist}) \cdot \left[\frac{L_{0ist}}{L_0} \right]^n \quad n = \text{empirical acceleration factor}$$

Accelerated Degradation Methodologies

Sum of lifetimes alternative empirical relation):

Example data: Green PHOLED

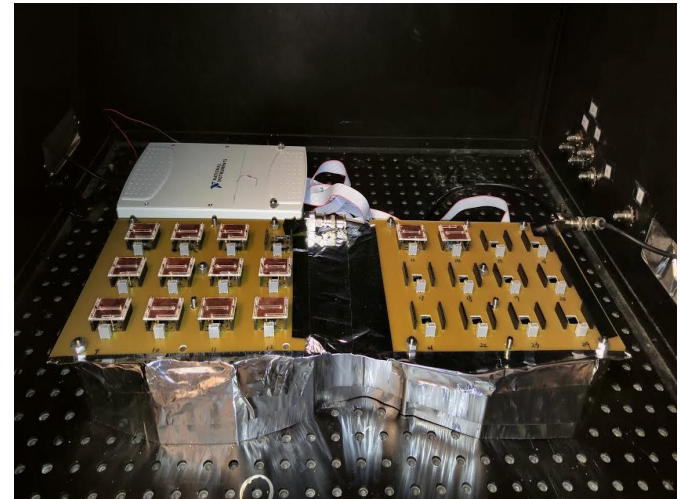


$$L(t)/L_0 = \lambda \exp(-t/\tau_1) + (1-\lambda) \exp(-t/\tau_2)$$

Burn-in
Long term decay

$$\frac{1}{\tau_2} = K'' j^\alpha \exp(-\Delta E_{A0}/k_B T)$$

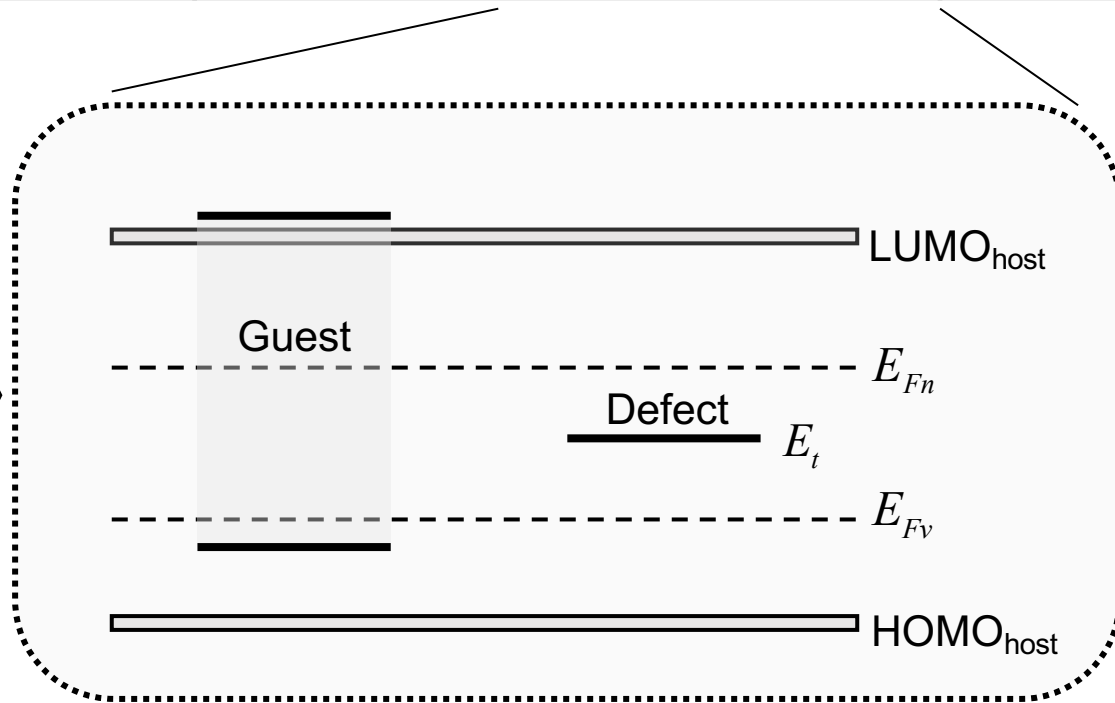
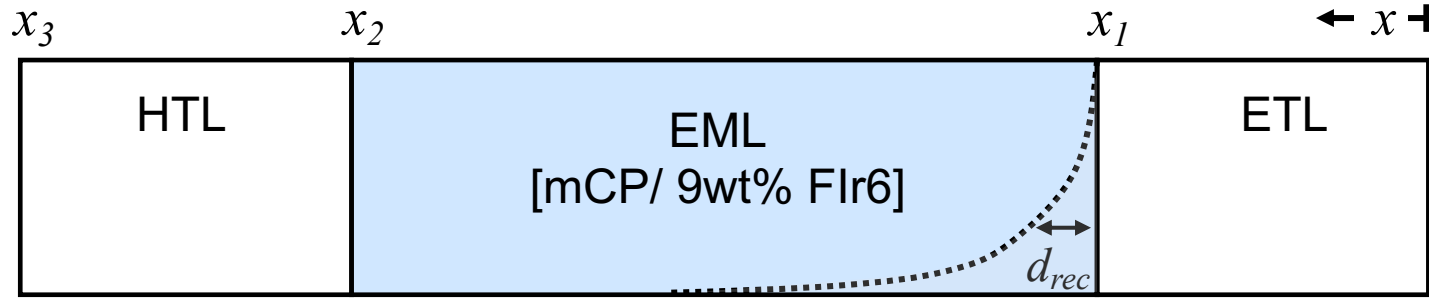
ΔE_{A0} = thermal activation of degradation
 α = current acceleration factor



Measuring populations of identical devices



Intrinsic Lifetime Limits of OLEDs

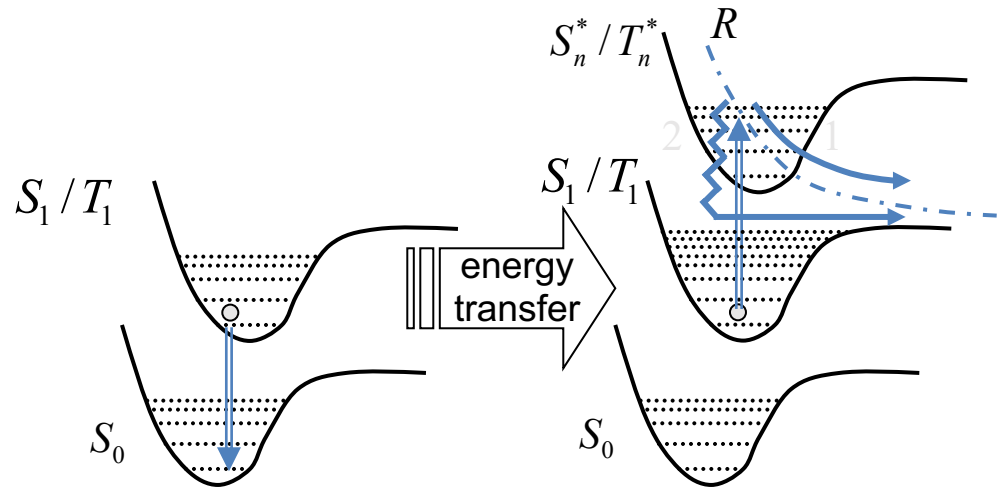


Energy Scale
 Red light: ~ 2 eV
 Green light: ~ 2.3 eV
 Blue light: ~ 2.9 eV

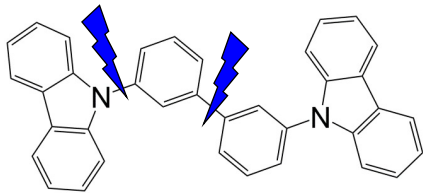


Degradation Routes

- Energetically Driven
 - Lifetime: $R > G > B$
- Two particle interactions lead to luminance loss
 - Exciton on phosphor, polaron on host
 - Exciton-exciton also possible



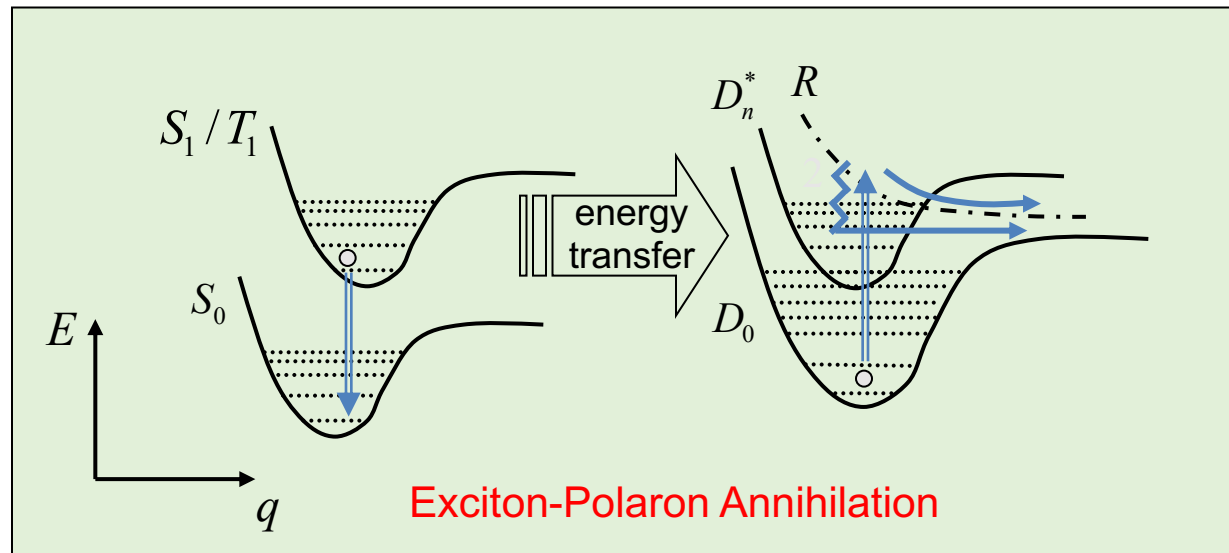
Exciton-Exciton Annihilation



Bond	BE(eV)	Bond	BE(eV)
C-C	3.64	N-N	1.69
C-H	4.28	N-O	2.08
C-O	3.71	N-H	4.05
C-N	3.04	O-O	1.51
C-F	5.03	H-H	4.52

Bond cleavage

Broken bonds? → Defects!

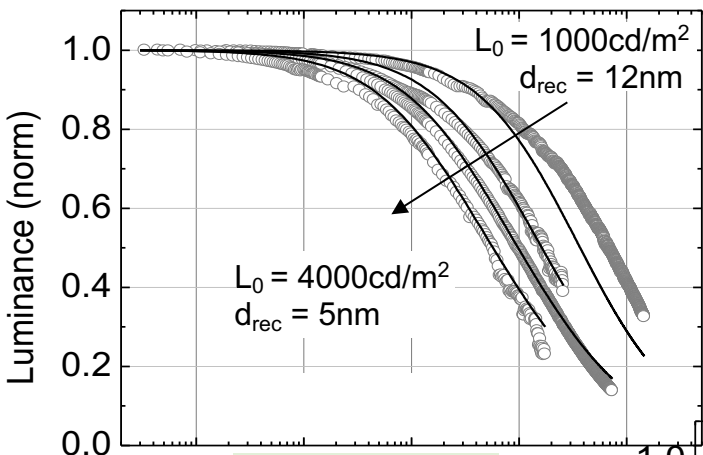


Exciton-Polaron Annihilation

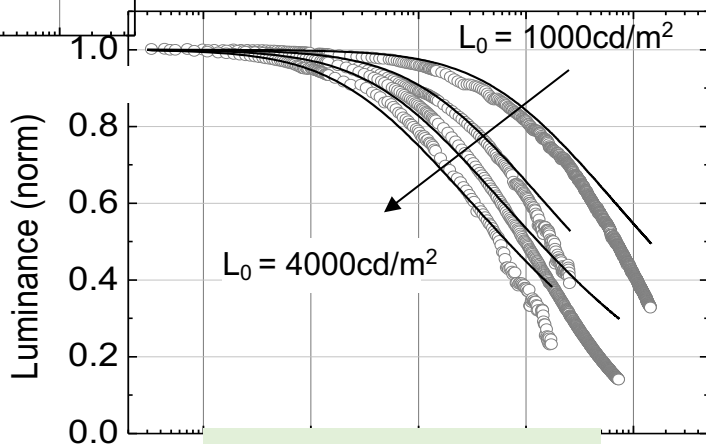
Triplet energy (~2.8 eV) + polaron (~3.3 eV) = hot polaron (≥ 6 eV)

Luminance Decay vs Time

- Blue PHOLED
- Prepared and packaged using industry std.

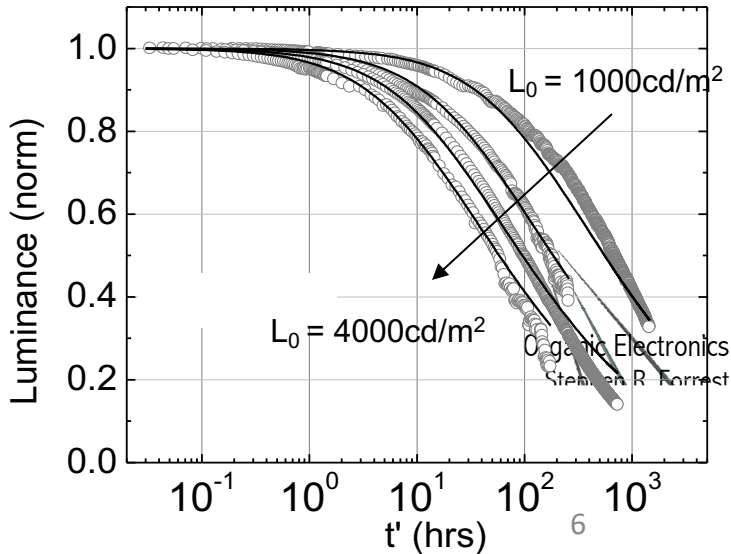


Exciton Localization



Exciton-Exciton Annihilation

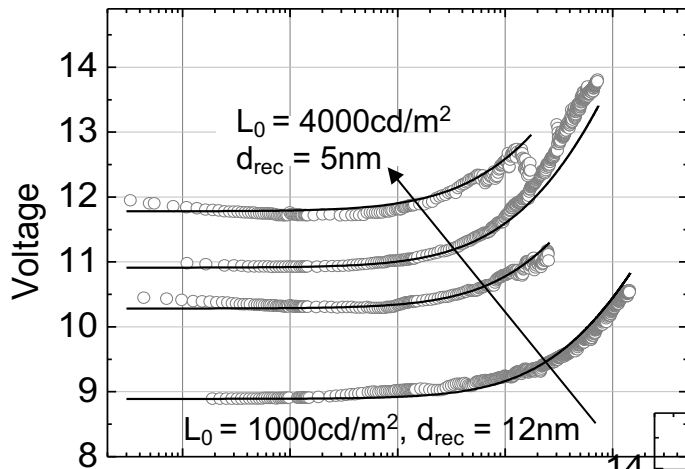
Exciton-Polaron Annihilation



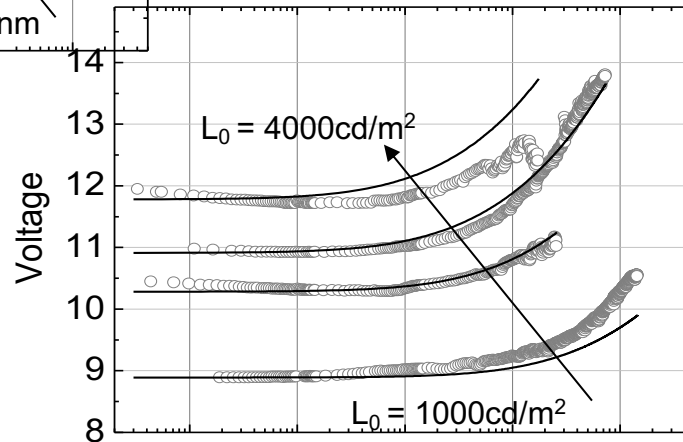
Defect Generation Rates

$$\frac{dQ(x,t')}{dt'} = \begin{cases} K_X n(x,t') & K_X p(x,t') & \text{P} \\ K_X N(x,t') & & \text{E} \\ K_X N^2(x,t') & & \text{E-E} \\ K_X N(x,t') n(x,t') & K_X N(x,t') p(x,t') & \text{E-P} \end{cases}$$

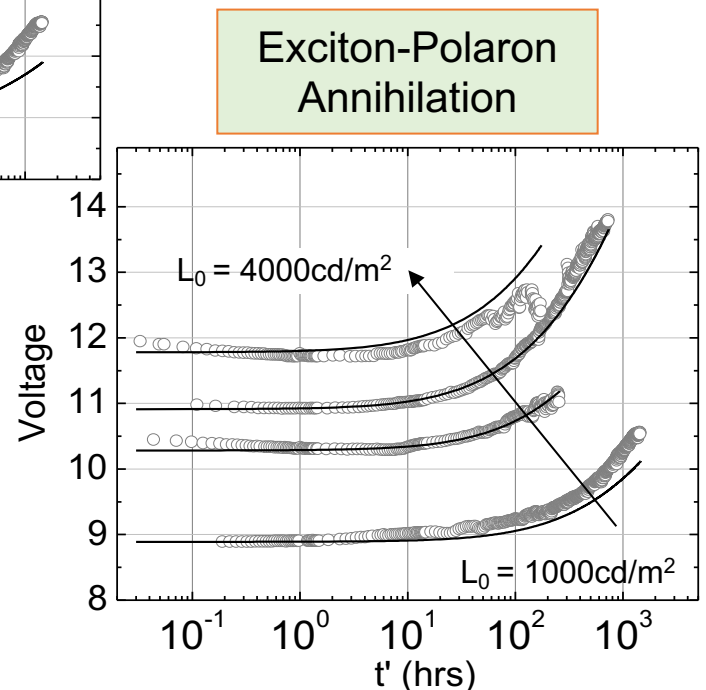
Drive Voltage Drift with Aging



Exciton
Localization



Exciton-Exciton
Annihilation

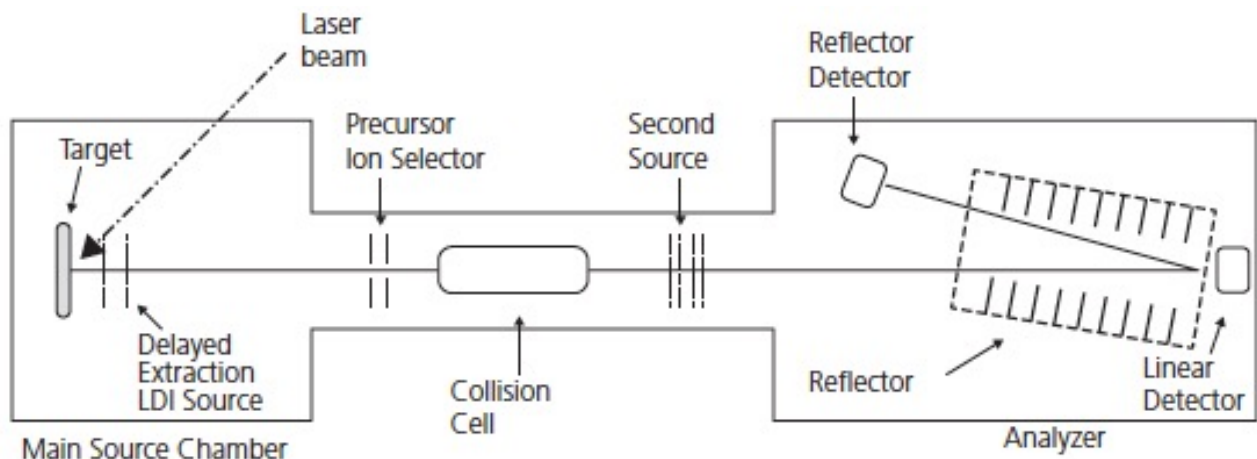


Exciton-Polaron
Annihilation

Conclusions

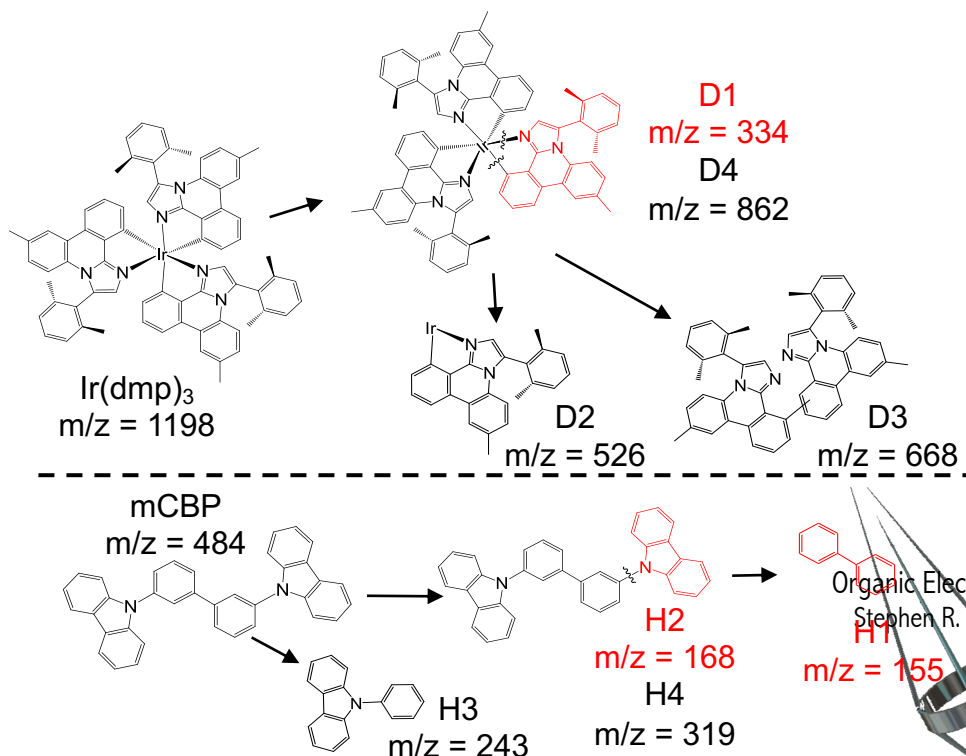
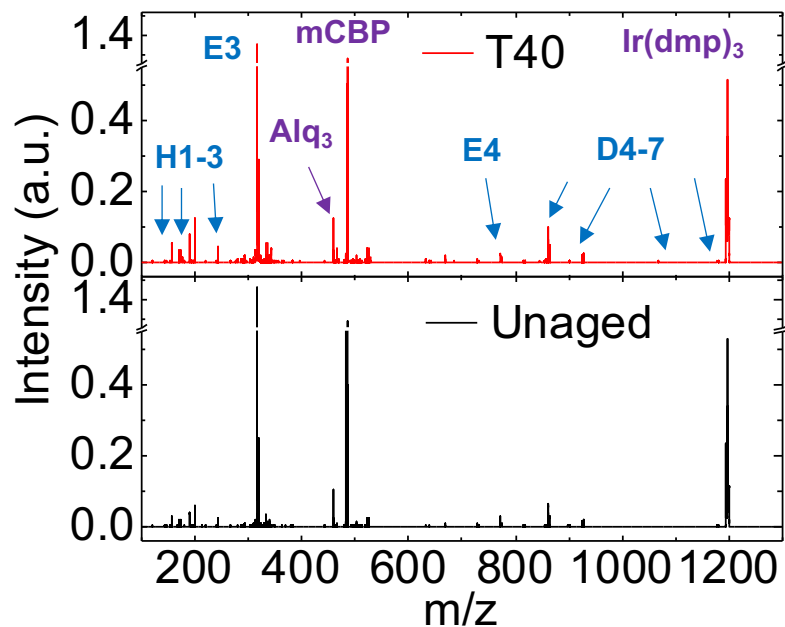
- $Q \sim 10^{18} \text{ cm}^{-3}$ → 50% increase in quenching
- At 1000 cd/m^2 , formation rate = $10^{12} \text{ cm}^{-2} \text{ s}^{-1}$
 - 1 in 5×10^8 E-P encounters leads to defect
 - Increasing recombination zone width extends lifetime
 - **Guest triplets/host polarons most active**

Evidence for Defect Formation: Molecular Fragmentation

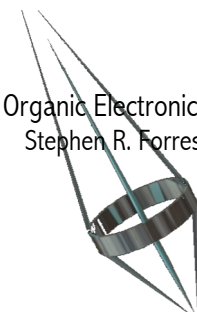
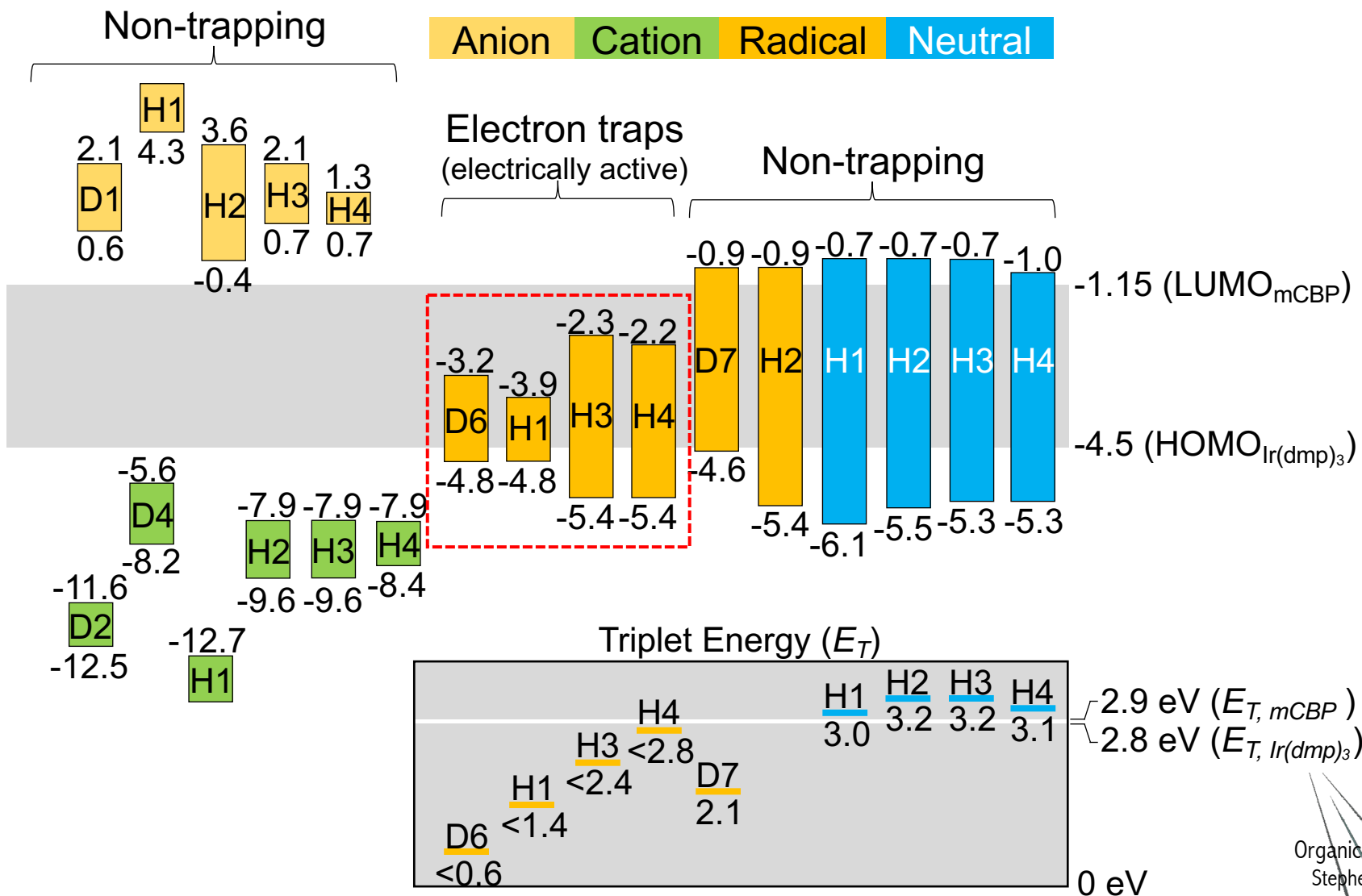


Laser Desorption Ionization-
Time of Flight Mass Spectroscopy
(LDI-TOF-MS)

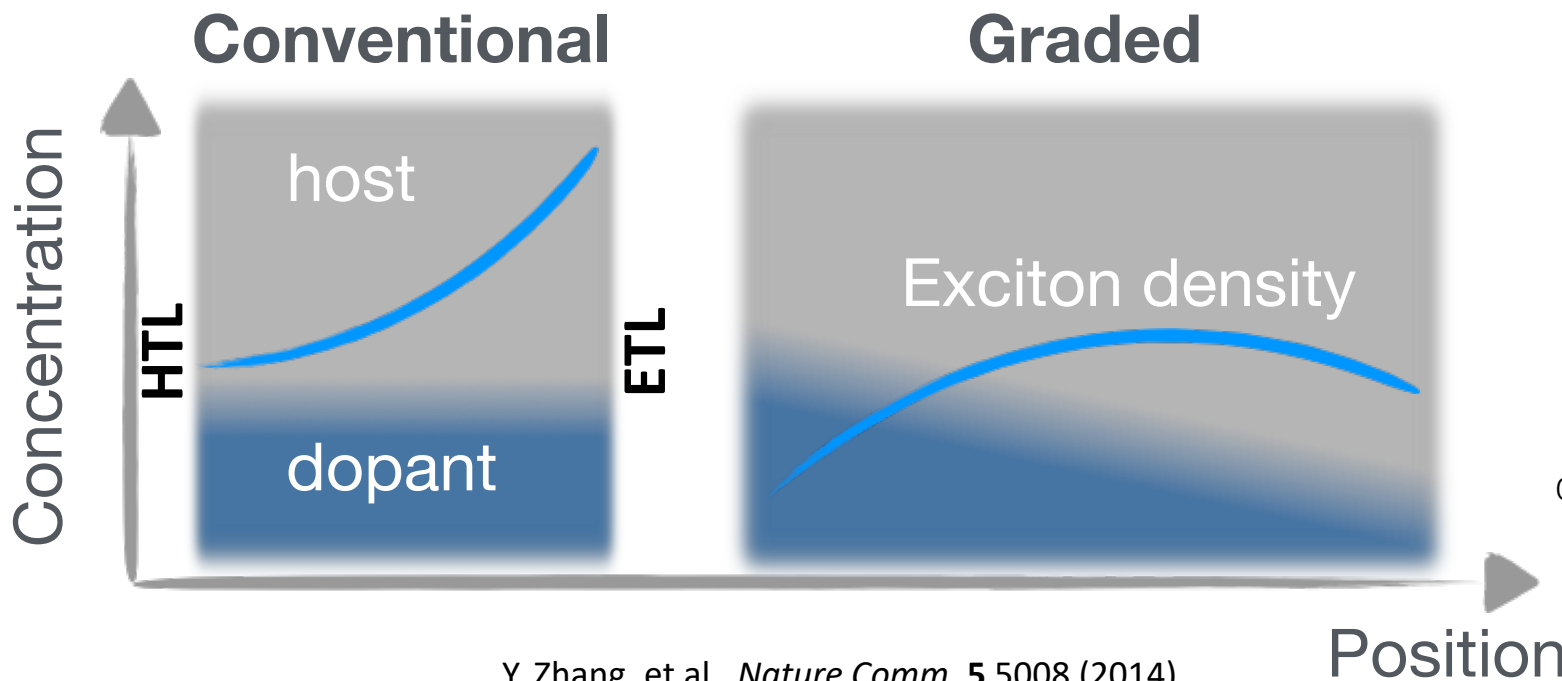
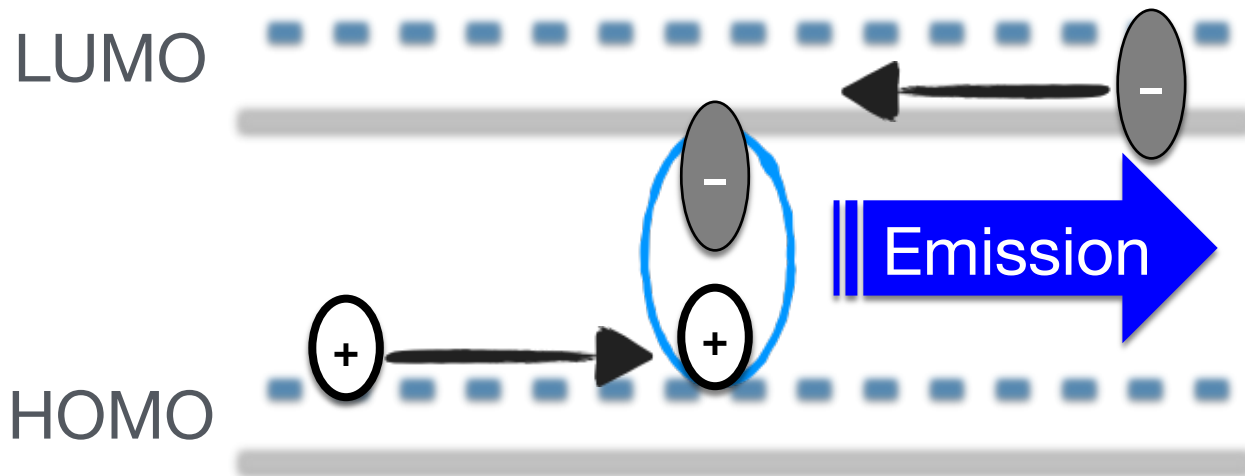
-Molecular species identification



Identification of Defect Energies

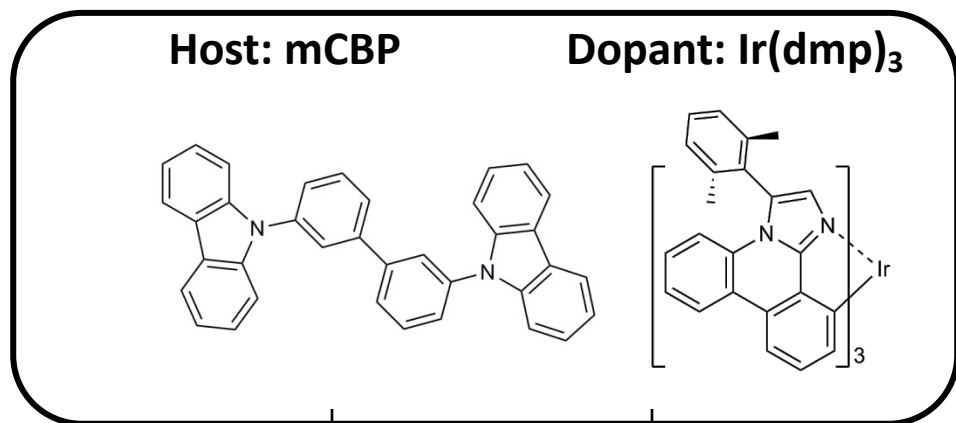


Reducing Exciton Density to Increase Lifetime



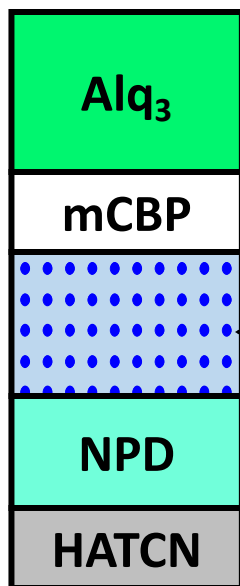
Spreading the recombination zone: Dopant/Host Grading

3 Different test device structures

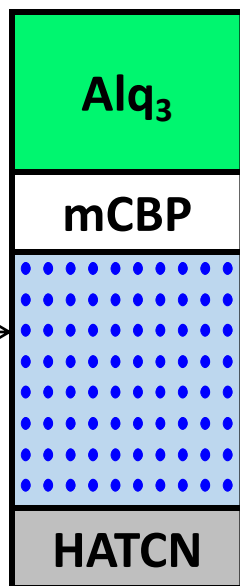


13 vol% uniform

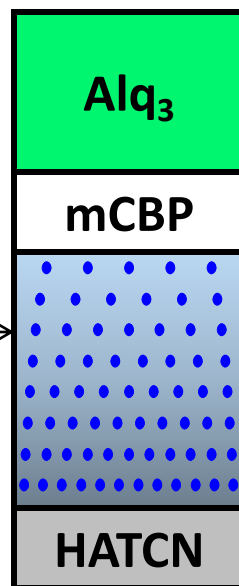
8 to 18% vol% graded



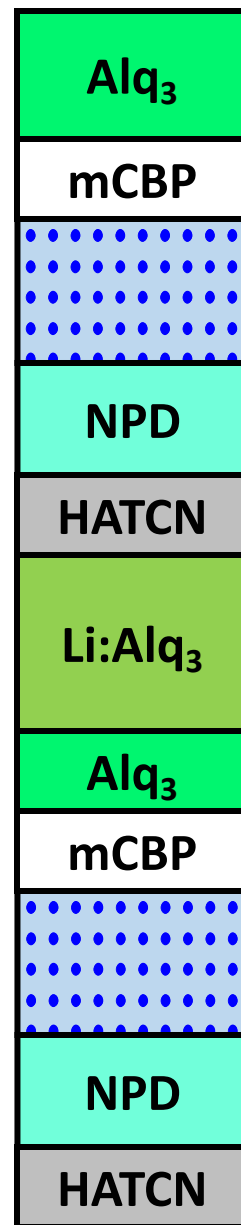
D1



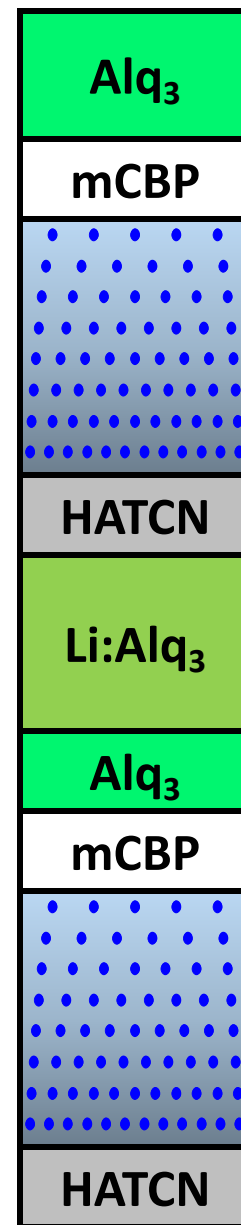
D2



D3



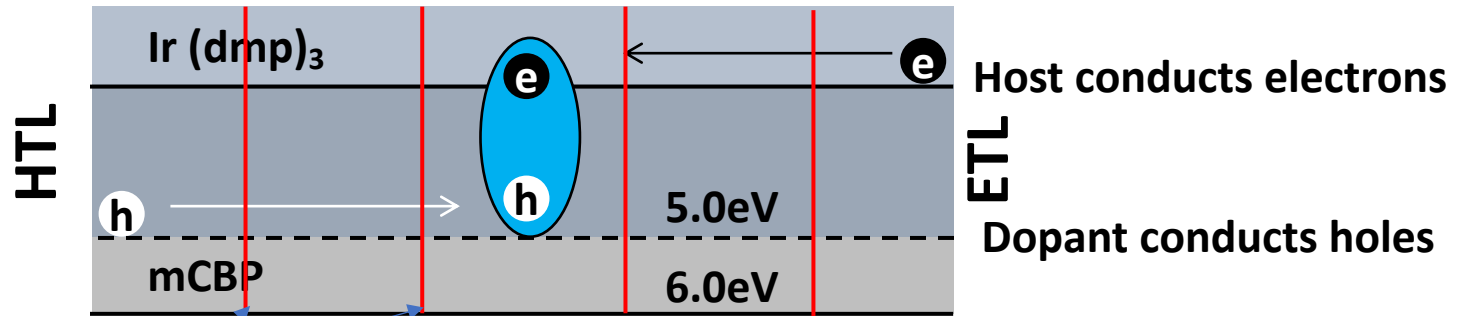
D1S



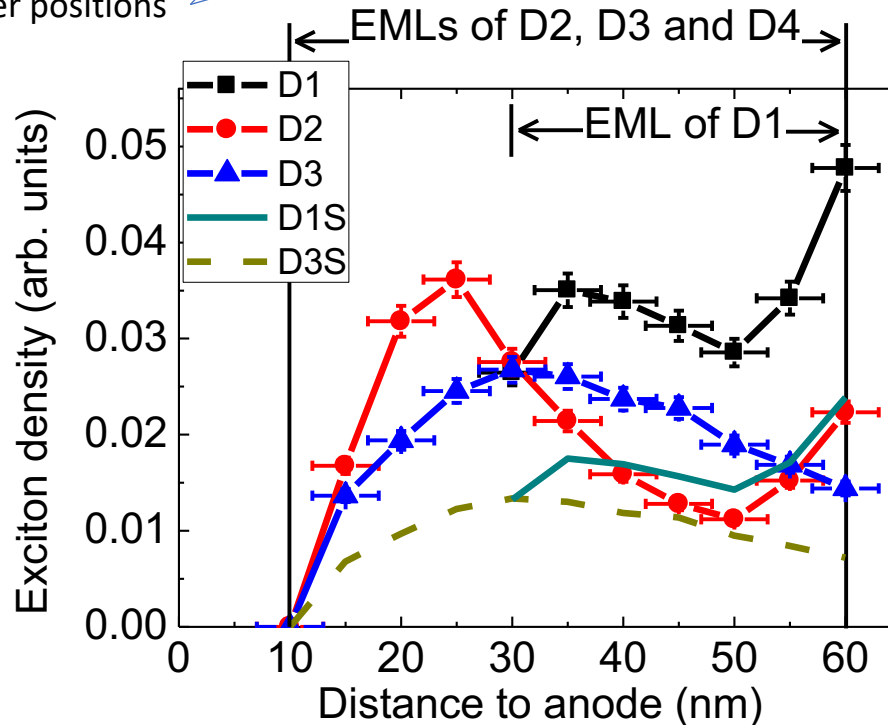
D3S



Excitons in the EML



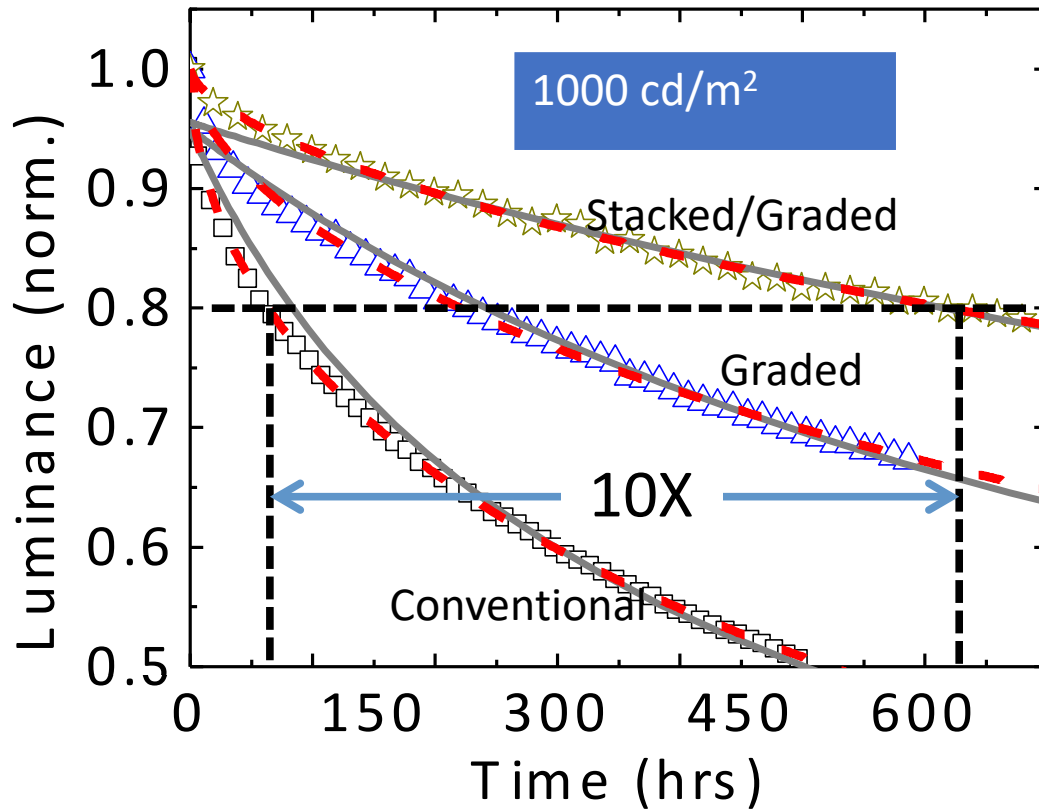
sensing layer positions



Exciton Sensing

- Red Phosphor (PQIr)
- 1.5 nm thick
- Placed at 5 nm intervals in EML
- Measure red emission intensity

10 X Lifetime Improvement Over Conventional



Y. Zhang, et al., *Nature Comm.* 5 5008 (2014)

Stacking is essential!



Panel 15 cm x 15 cm 82% fill factor	2 Unit WSOLED
Luminance [cd/m ²]	3,000
Efficacy [lm/W]	48
CRI	86
Luminous Emittance [lm/m ²]	7,740
1931 CIE	(0.454, 0.426)
LT₇₀ [hrs]	13,000

P.Levermore et al, *SID Digest*, 2011.

Dopant Grading: Is it Good Enough?

using acceleration factors to predict lifetime

- Luminance to achieve sRGB color gamut for G is 10X that for B
- \Rightarrow B sub-pixel $L_0=100 \text{ cd/m}^2$ (c.f. G with $L_0>1,000 \text{ cd/m}^2$)
- \Rightarrow B lifetime to T50=70,000 hr.
- Adopting Degradation acceleration factor: $n = 1.55$ with

$$T50(100\text{cd/m}^2) = T50(1000\text{cd/m}^2) \times \left[\frac{1000\text{cd/m}^2}{100\text{cd/m}^2} \right]^n$$

- \Rightarrow B PHOLED lifetime to T50 = 1.3×10^5 hr.
- Commercial G PHOLED lifetime = 10^6 hours at $L_0 = 1000 \text{ cd/m}^2$.

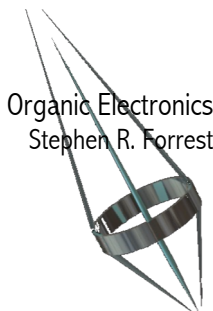
Not blue enough, T95 is required



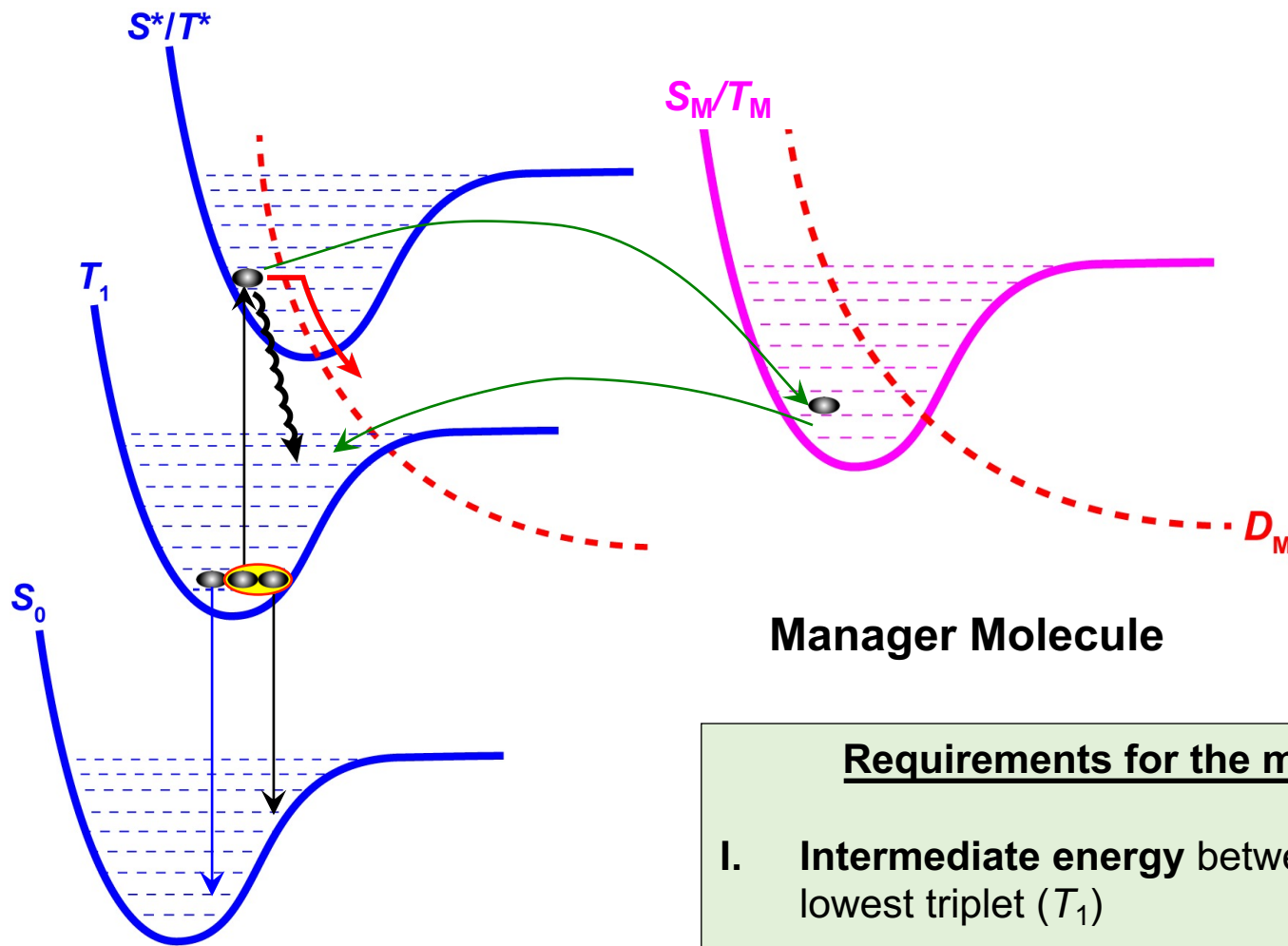
Dopant Grading for Lighting: Is it OK?

- Current state of stacked WOLED: T70=13,000 hrs
- Mostly limited by blue lifetime
- Only light blue required
- Estimated increase in lifetime for stacked blue at lighting brightness: ~4X
- Lifetime of blue lighting using grading: 50,000 hr

This is almost good enough



Hot excited state management: Eliminating the highest energy excited states



Blue Dopant/Host Molecules

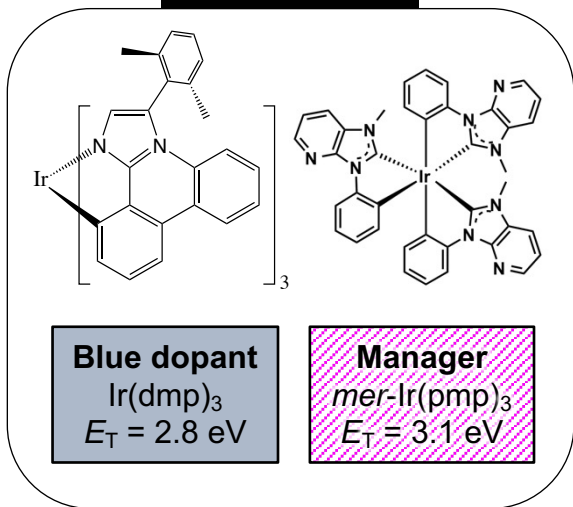
Manager Molecule

Requirements for the manager molecule

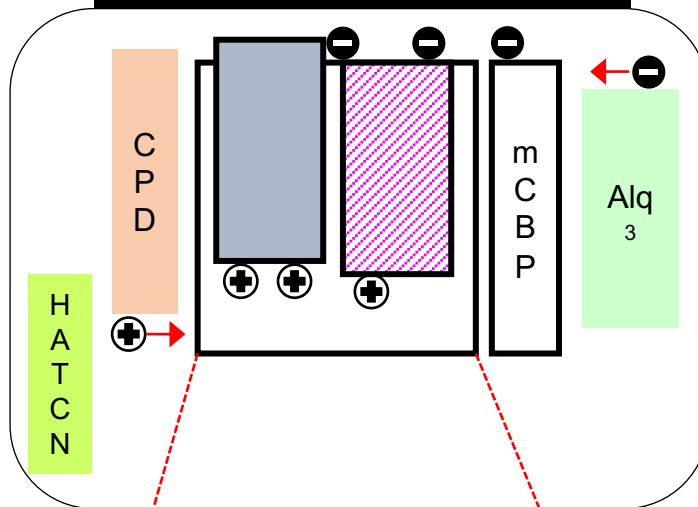
- I. **Intermediate energy** between hot state (T^*) and lowest triplet (T_1)
- II. **Molecular stability**
- III. **Fast energy transfer** from dopant/host to manager

Managed blue PHOLEDs

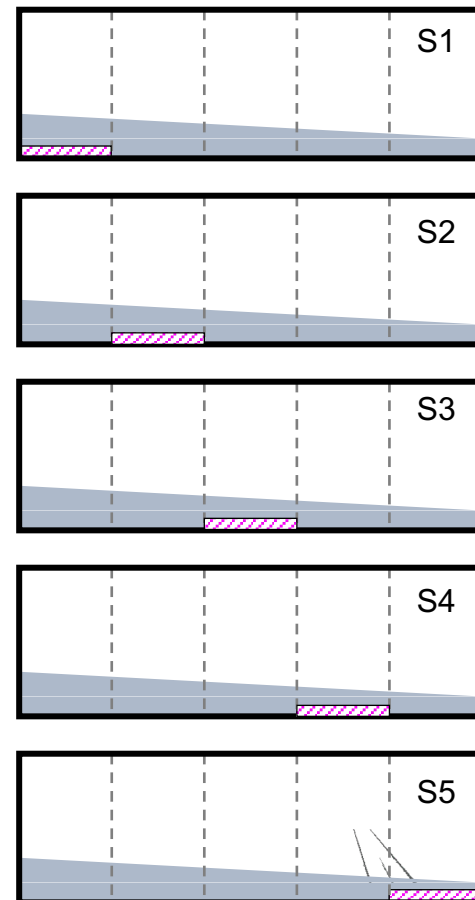
EML materials



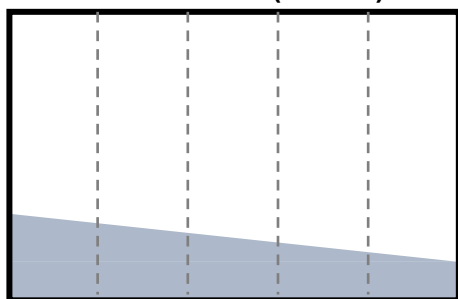
Energetics and charge transport



Managed EML (M1–M5)

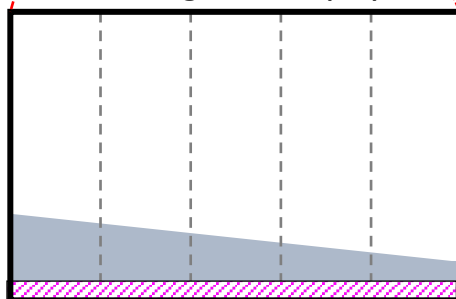


Graded EML (GRAD)



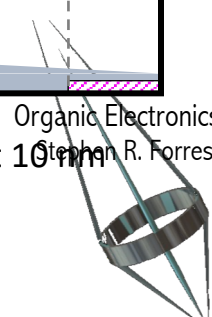
■ : 18–8 vol%

Managed EML (S0)

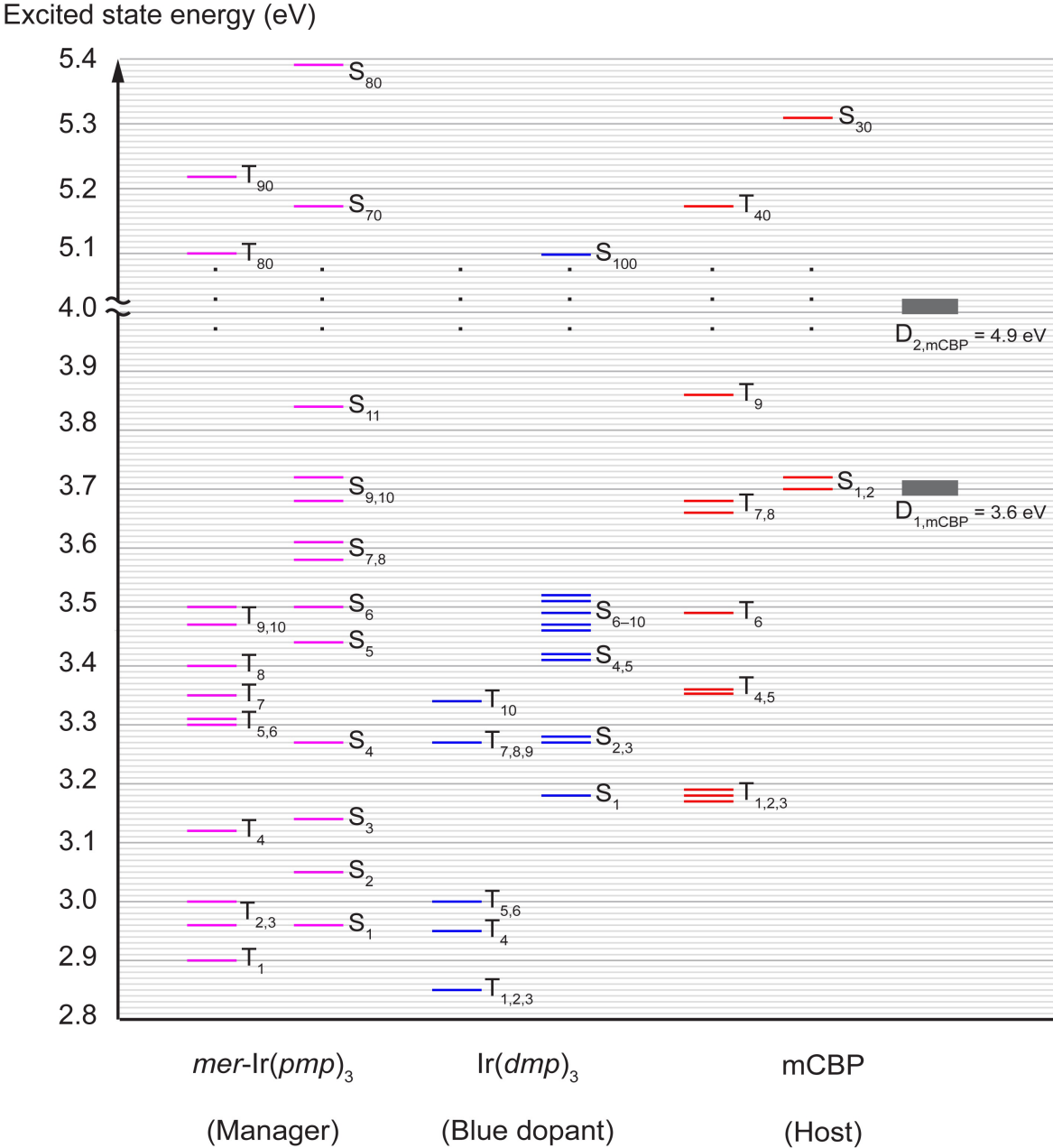


■ : 15–5 vol%
 ■ : 3 vol%

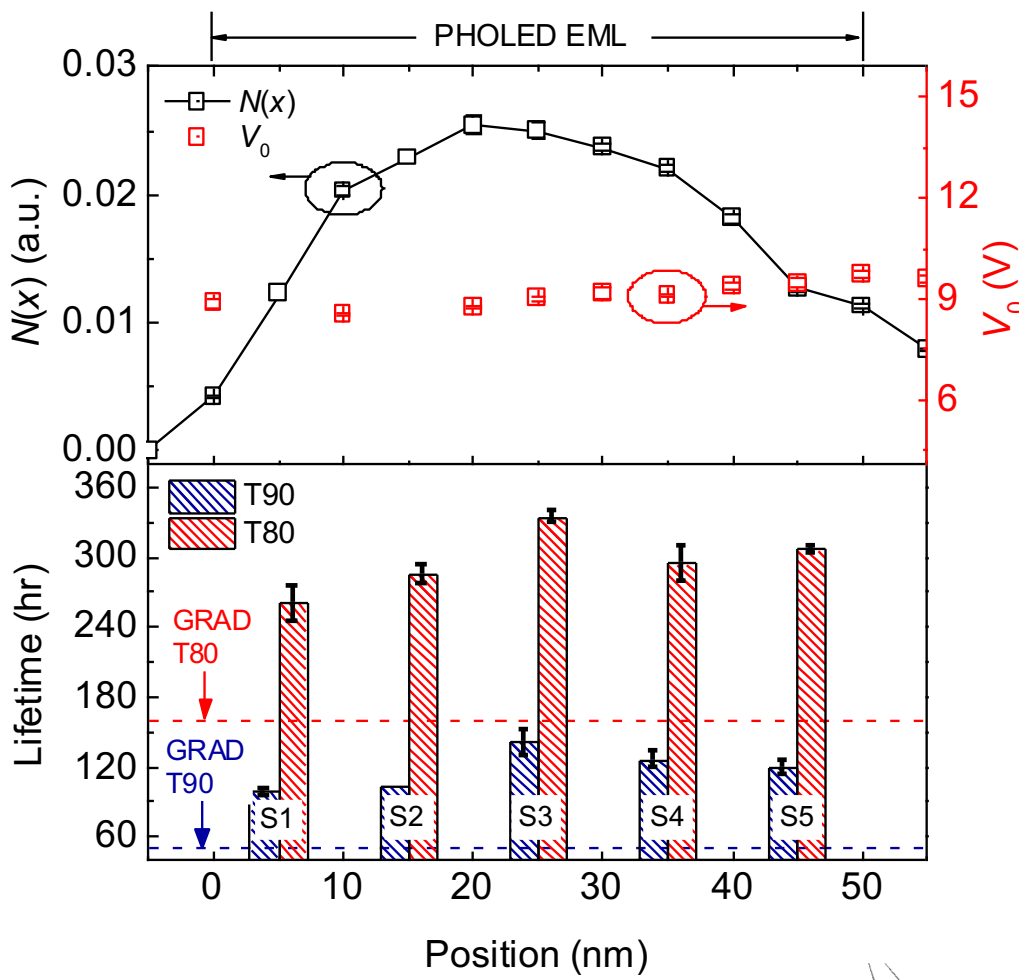
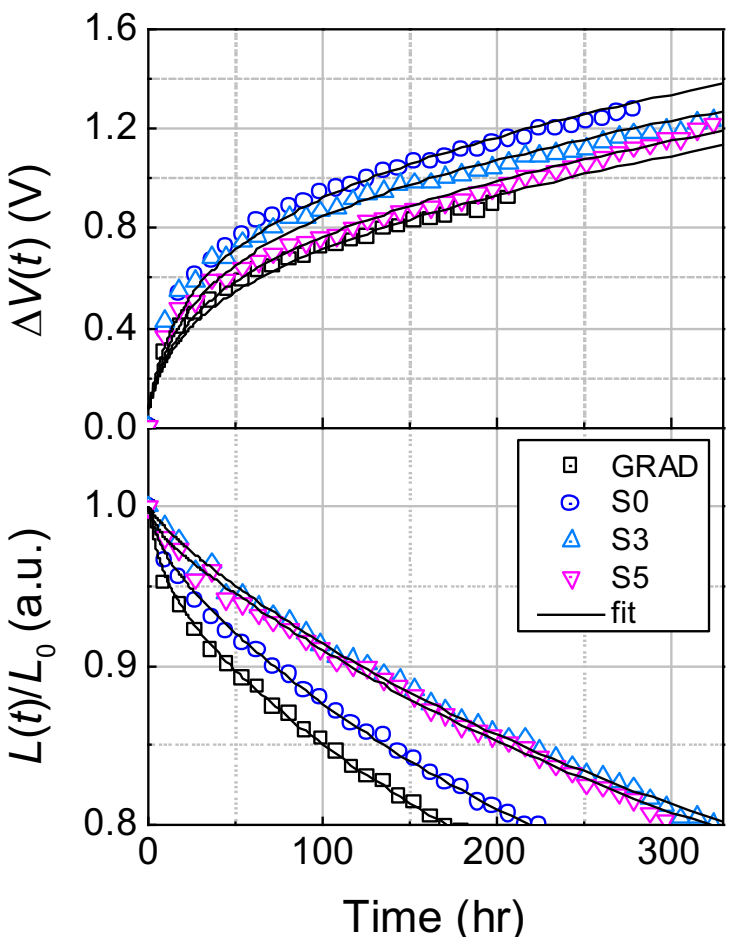
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 Place manager at 10 nm sections of EML



Plenty of Energy Levels to Access in the Management Process

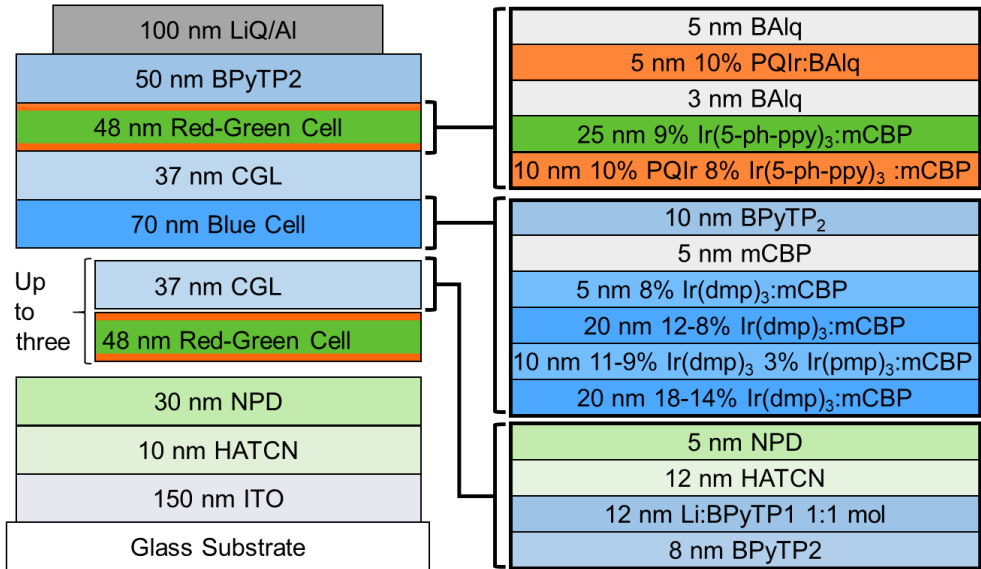


Lifetime Improvements and TTA/TPA Model



- Greatest improvement in lifetime when manager at position of highest exciton density (S3)
- Fractional increase in lifetime decreases with time
 - Greater at T90 than T80 ⇒ manager depletion

Putting Grading Excited State Management to Work: Long lived all phosphor stacked WOLEDs



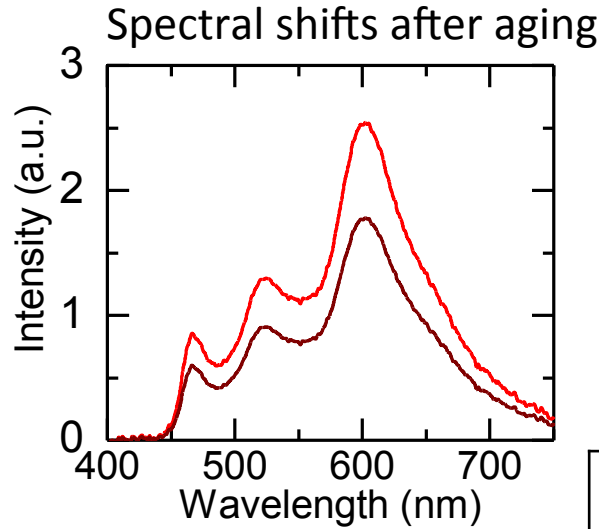
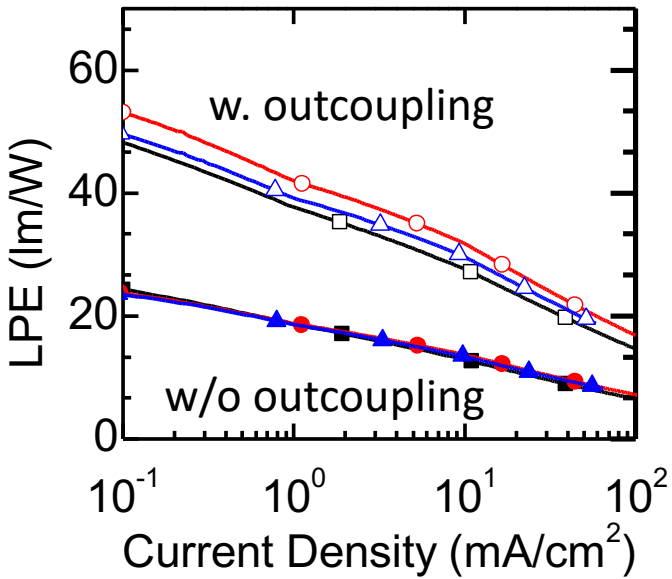
- Max Luminance > 200,000 nits
- 50 lm/W max
- CCT = 2780K
- CRI=89



Photo illustrating good color rendering of the SWOLEDs in this report. The luminaire comprises 36 pixels (2 mm²) operated at 50-100k nits



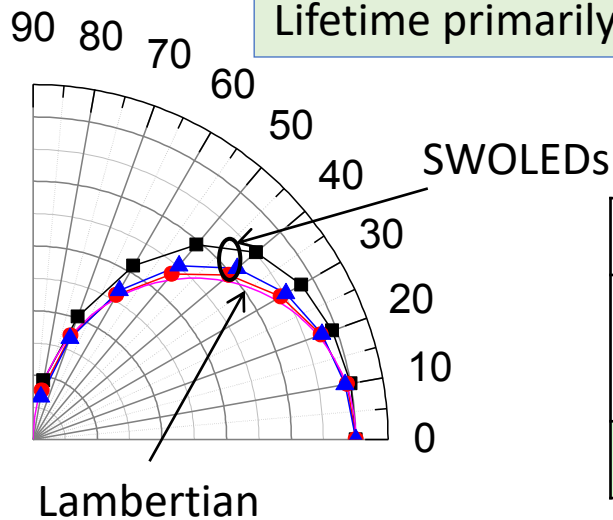
All Phosphor SWOLED Performance



T70	SWOLED
Δ CCT	-360 K
Δ CRI	-0.8
Δ CIE	(0.03,0)

SWOLED Architecture	Blue degradation @ WOLED T70:
Conv	T28
Grad-Managed	T48

Lifetime primarily limited by R/G sections



	With outcoupling		$\Delta V/V_0$ (T70) (%)
	T70 1000 nit (x10 ³ hr)	T70 3000 nit (x10 ³ hr)	
SWOLED	80±40	14±5	~+10%

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What we learned about OLEDs

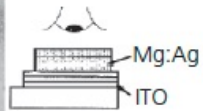
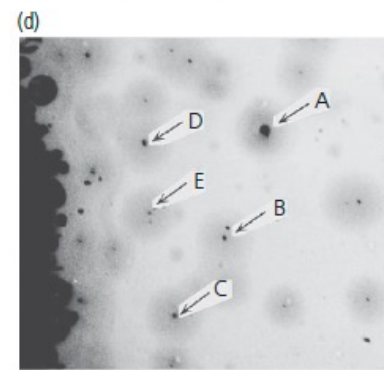
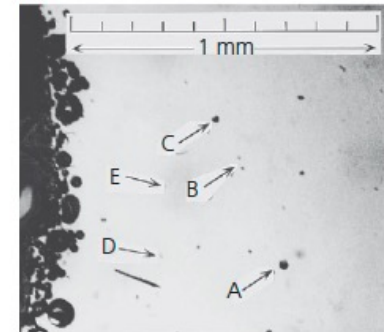
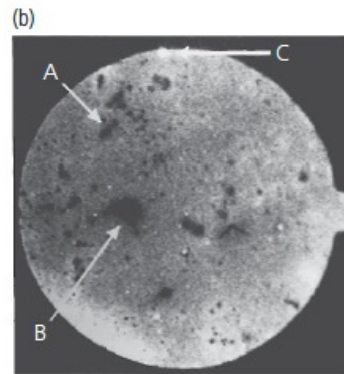
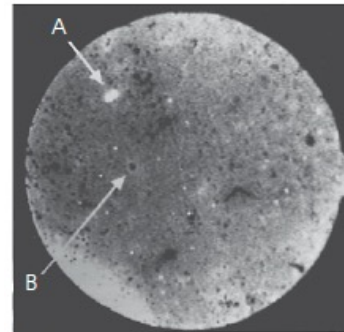
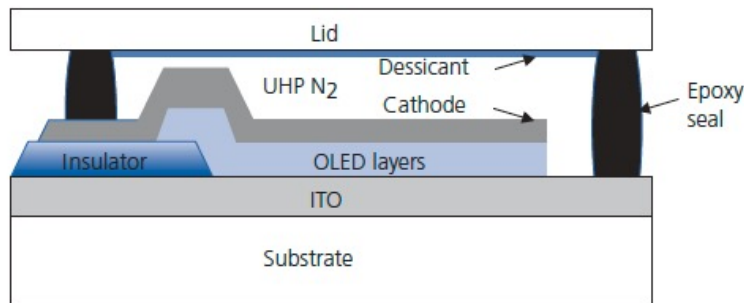
- Chromaticity and the perception of color is quantified based on eye response (photometric quantities)
- OLEDs reach highest efficiency when both singlets and triplets are harvested (heavy metal complexes and TADF molecules)
- Optimized OLEDs have many layers serving purposes ranging from charge conduction, contacting to electrodes, to light emission
- Outcoupling methods essential to view substrate and waveguide modes while limiting surface plasmons
- Degradation of OLEDs particularly severe for blue due to bimolecular annihilation
- Lighting requires broad spectral emission using multilayer devices or excimer emission
- OLEDs provide uniform, area lighting vs. specular LED lighting



Packaging Matters

(see Ch. 5)

- Without packaging, there is rapid degradation of OLED luminance
- Dark spot defect formation of contacts more rapid in atmosphere than when packaged in inert (e.g. N₂) gas.



- Dark spot formation on cathode as seen from the top and bottom sides
- Defect bright EL spots soon become dark
- Defects appear to be due to dust on substrate penetrating device active region

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Organic Lasers

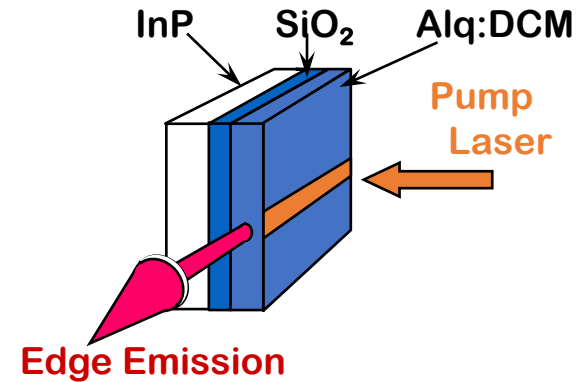
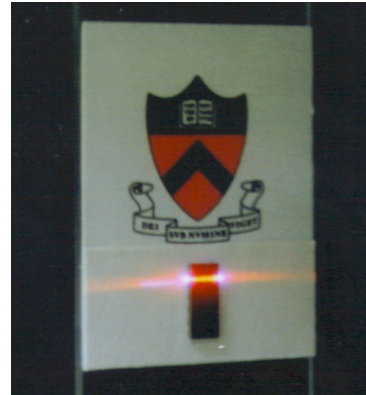
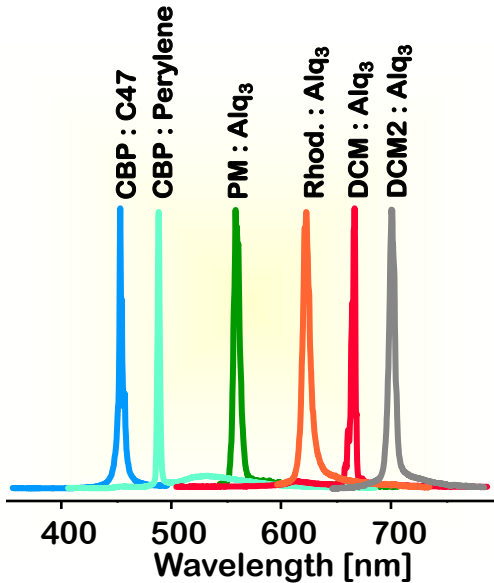
- High intensity, monochromatic sources
- To date, only optical pumping has unambiguously shown lasing
- Large triplet and electrode losses, and low mobilities (hence low current) hamper electrical pumping

How to identify a laser

- A clear threshold between spontaneous and stimulated emission evident from an abrupt increase in output slope efficiency
- A significant narrowing of the spectral linewidth at threshold. Single mode lasers show linewidths $\sim 1\text{\AA}$, multimode lasers will have multiple emission lines coincident with the gain (PL) spectrum of the material, and whose separation is $\Delta\lambda \sim 1/L$, the cavity length
- A well-defined output beam
- Temporal and spatial coherence of the output beam

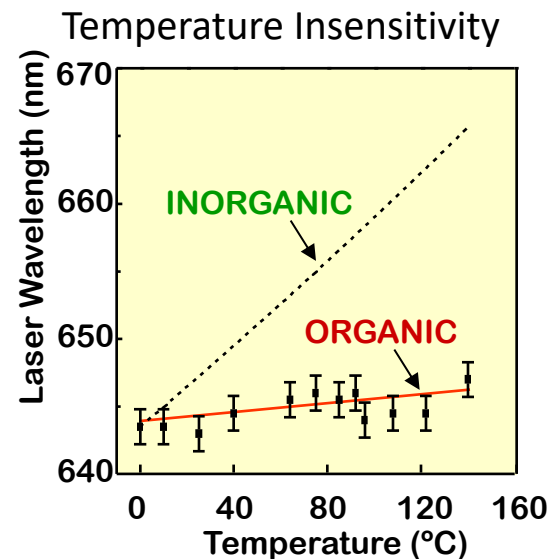
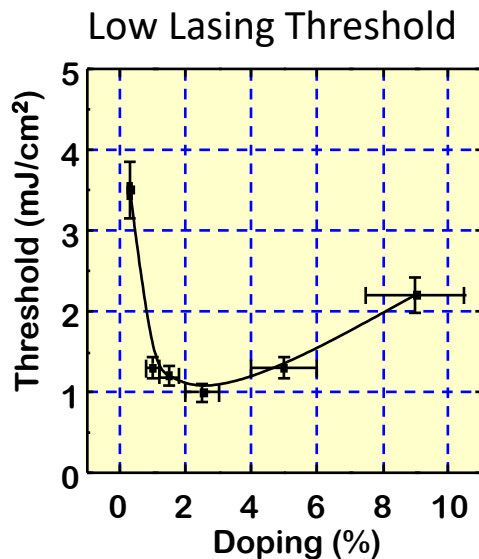


Features of Organic Lasers



Edge Emission
Kozlov, et al., *Nature* **389**, 362 (1997).

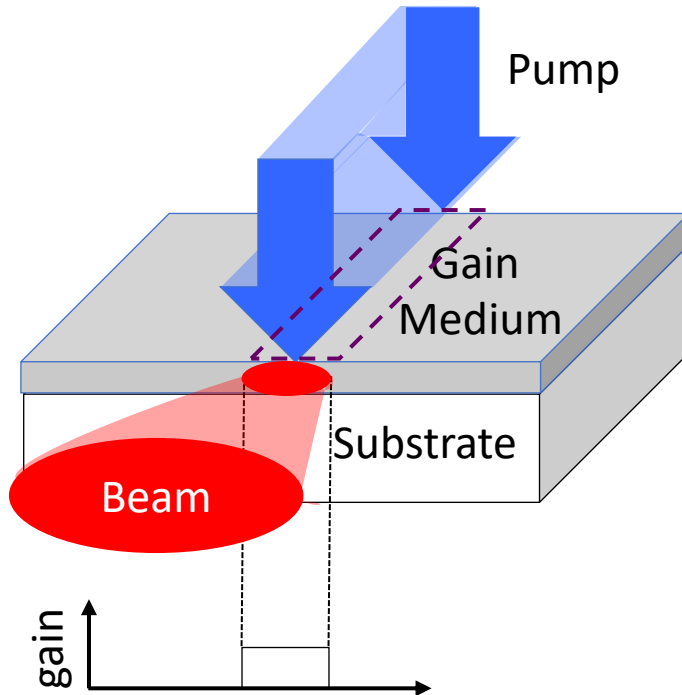
- Material Tunability
- Freedom from Epitaxial Limitations
- Natural Quantum Dots



Kozlov, et al., *Appl. Phys. Lett.* **71**, 2575 (1997).

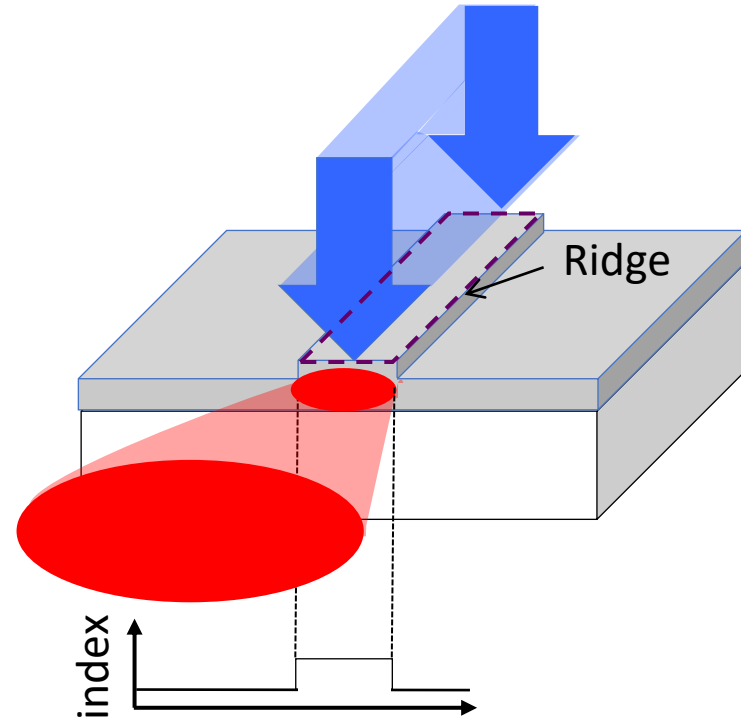


Gain vs. Index Guided Lasers



Gain guiding

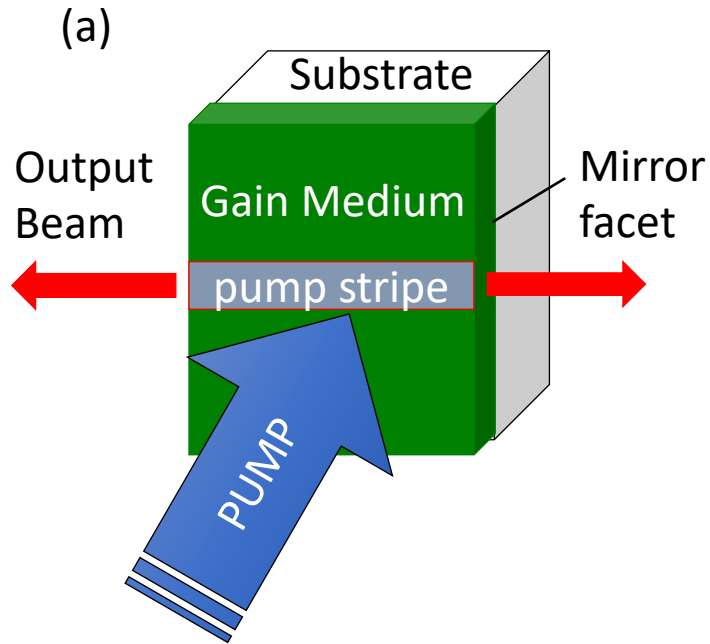
- Simple
- Can lead to high thresholds
- Can lead to modal instabilities



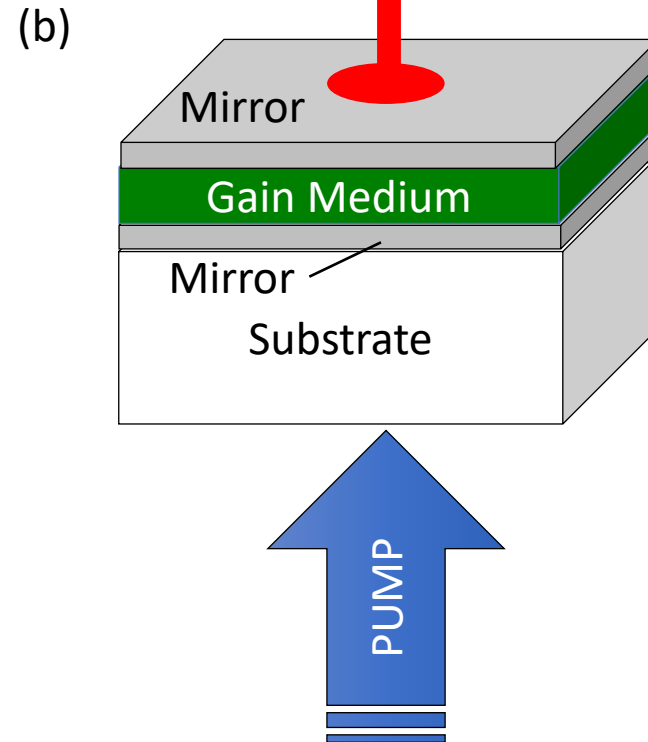
Index guiding

- More complex
- Can reduce thresholds
- Has modal instabilities only at very high power

Optically pumped lasers



Longitudinal Configuration

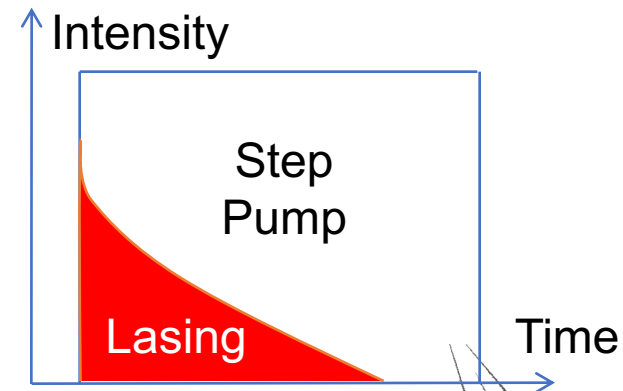
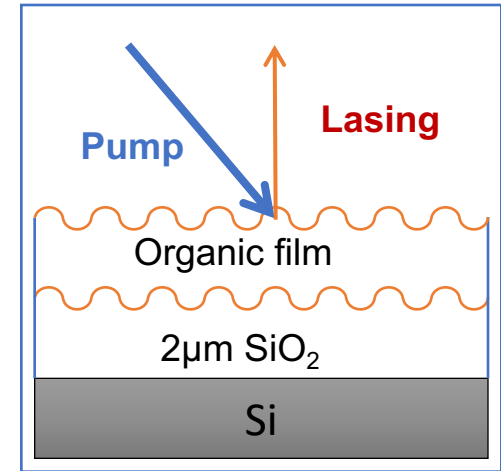


Vertical Cavity Surface Emitting Laser (VCSEL)



Pulsed Organic Lasers

- Why does organic lasing last only $<100\text{ns}$?
 - Initially ($<10\text{ns}$)
 - Negligible T
 - Gain=Loss
 - Later ($>100\text{ns}$)
 - T builds up
 - Gain \downarrow : S-T quenching
 - Loss \uparrow : T absorption
 - Same source of loss prevents electrically pumped laser action



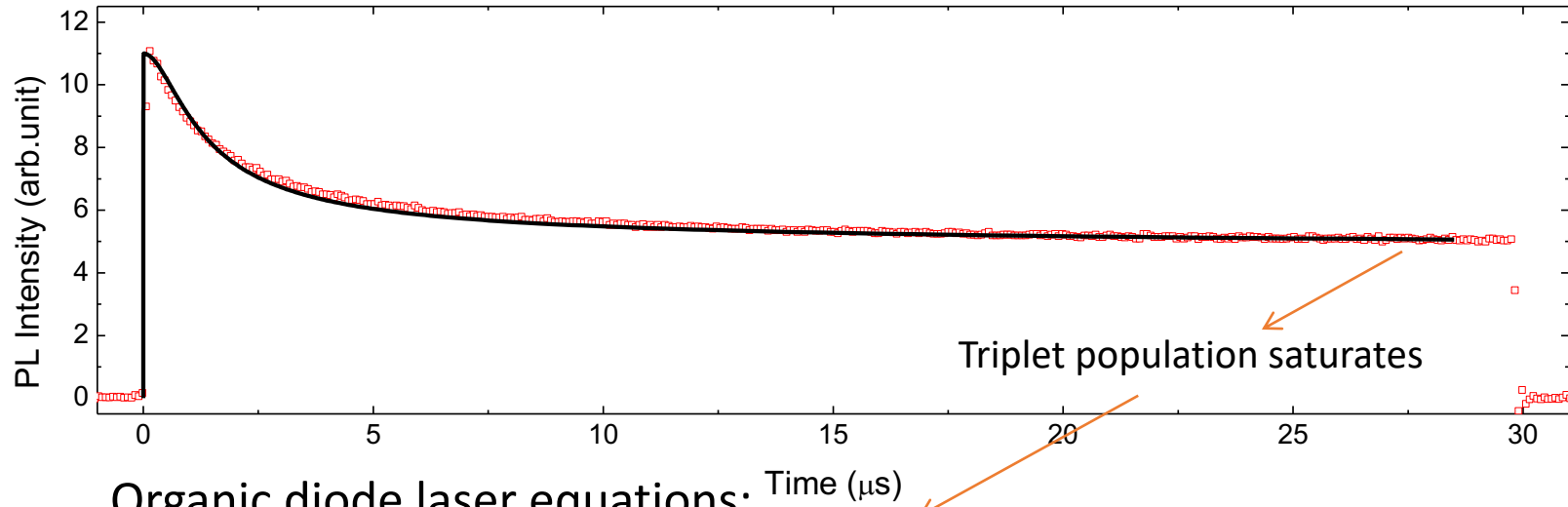
Giebink & Forrest, *Phys. Rev. B* **2009**, 79, 073302

Lehnhardt, et al. *Phys. Rev. B* **2010**, 81, 165206

Organic Electronics
Stephen R. Forrest



Triplet saturation and CW threshold



Organic diode laser equations:

Threshold: $g_{net}(t) = \Gamma \sigma_{stim} S(t) - \alpha_{cav} - \Gamma \sigma_{TT} T_G(t) \geq 0$ (gain condition)

Pulsed threshold

$$I_{PS} = e_p d (k_S + k_{ISC}) \frac{\alpha_{CAV}}{\eta \Gamma \sigma_{stim}}$$

CW threshold

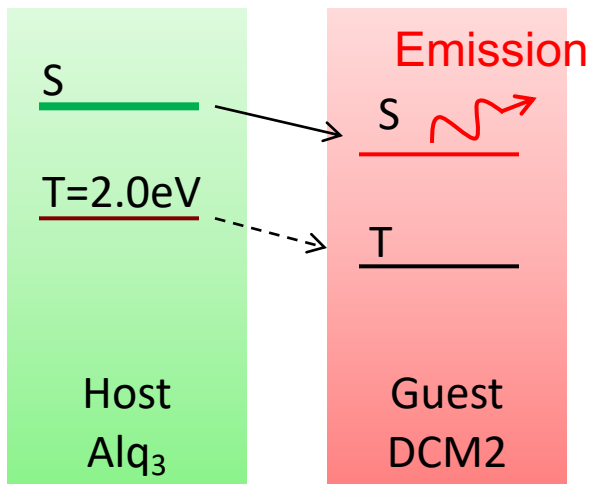
$$I_{CW} = e_p d (k_S + k_{ISC} + k_{ST} T_\infty) \frac{\alpha_{CAV} + \Gamma \sigma_{TT} T_\infty}{\eta \Gamma \sigma_{stim}}$$

S-T T abs

	I_{PS} (kW/cm ²)	T_∞ (10 ¹⁸ cm ⁻³)	I_{CW} (kW/cm ²)
DCM2:Alq ₃	0.93	5.0	32

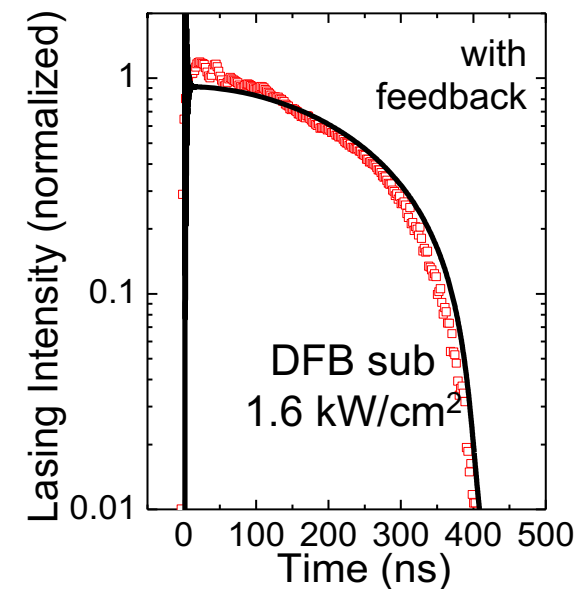
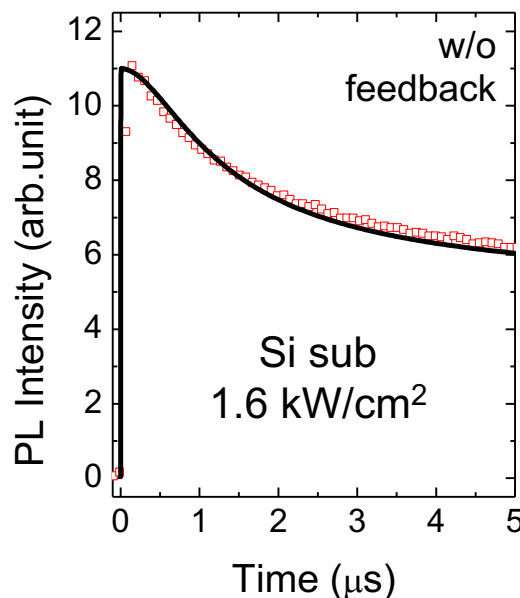
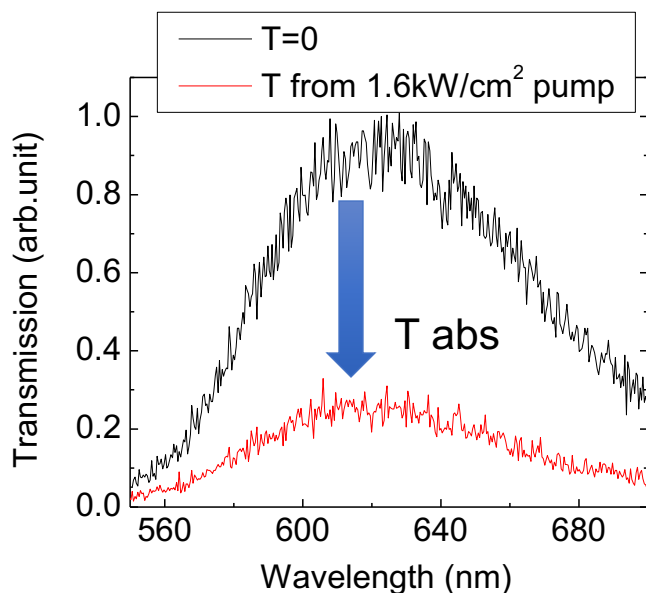
Above damage threshold

Example Laser -- DCM2:Alq₃



- PL fit follows S-T quenching
 - $S^* + T^* \rightarrow S_0 + T^{**}$
- Lasing fit follows S-T and T absorption
 - $P + T^* \rightarrow T^{**}$
- Lasing condition

$$g_{net}(t) = \Gamma \sigma_{stim} S(t) - \alpha_{cav} - \Gamma \sigma_{TT} T_G(t) \geq 0$$

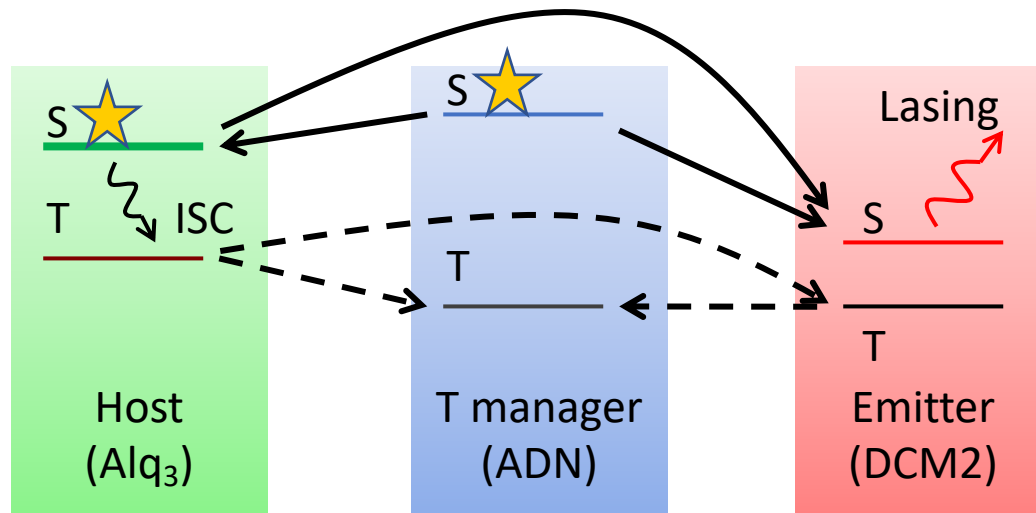


Lasing turns off once triplet loss exceeds gain



The Triplet Management Concept

- Introduce a 3rd molecule into the gain region that quickly removes triplets \Rightarrow reduced absorption loss
- Same concept works in fluorescent OLEDs to reduce rolloff due to accumulation of triplets at high intensity

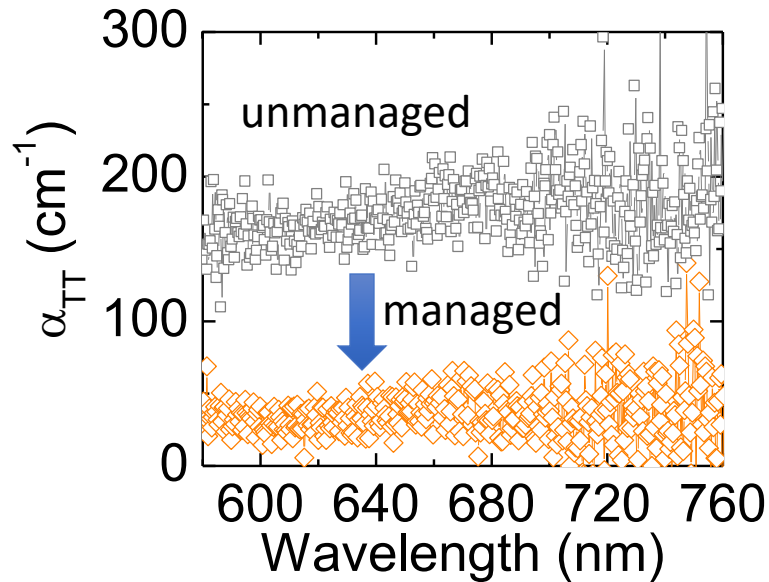


\longrightarrow Förster transfer \dashrightarrow Dexter transfer

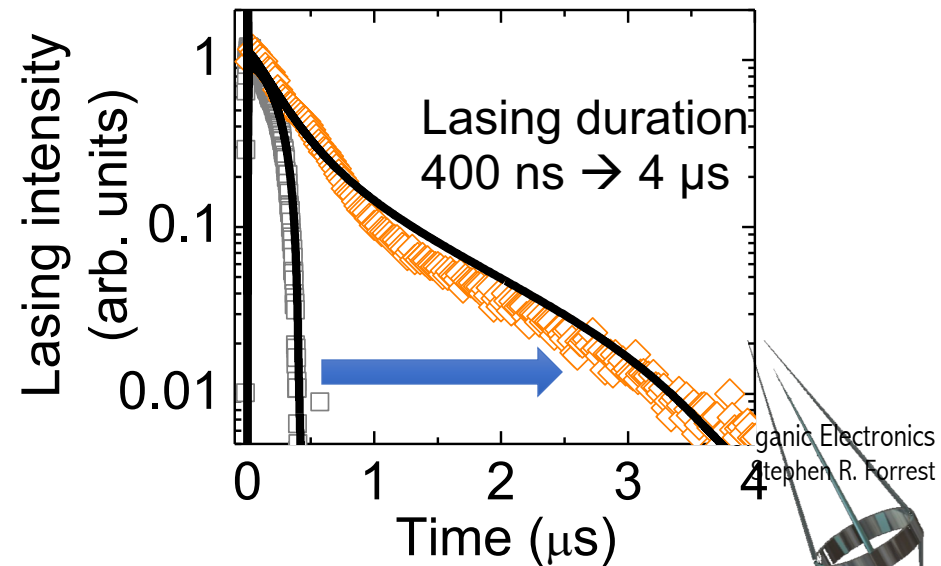
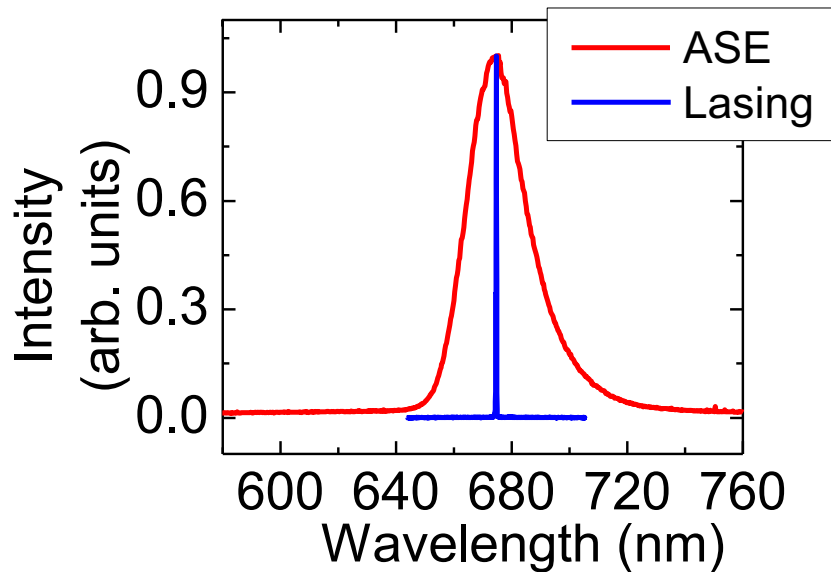
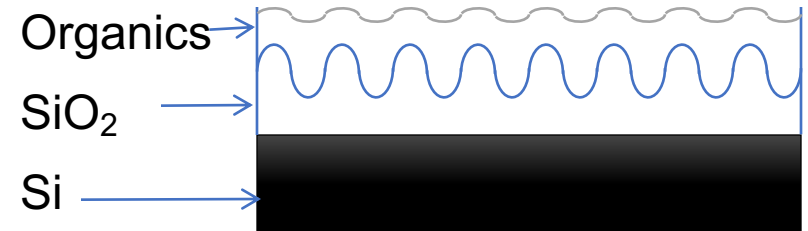
Example gain region composition

T absorption and lasing with T management

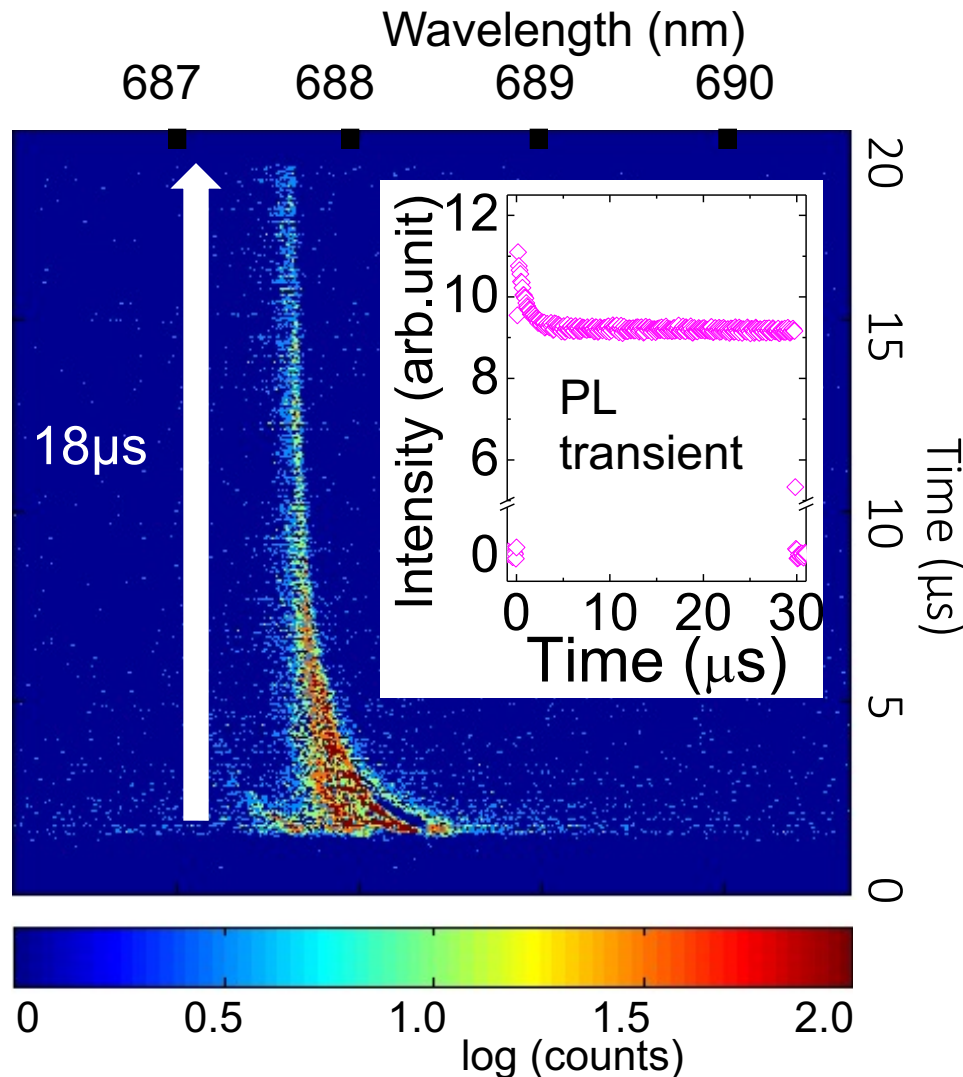
Alq₃:DCM2



Organic laser with 2nd order distributed feedback structure



Management Allows CW Lasing Above the CW threshold



□ Condition

- 2.4 kW/cm², 10 Hz/18 μ s
- Integrate 1000 pulses
- Degradation limited

□ **Single pulse \rightarrow 100 μ s**

□ Wavelength shift due to triplet induced index change

$$m\lambda = 2n_{\text{eff}}\Lambda$$

□ CW lasing not limited by T



What we have learned

- OLEDs are the leading application of organic electronics due to their features of:
 - Color versatility due to chemical modification
 - 100% internal efficiency in PHOLEDs and TADF molecules
 - Stability over long term operation (except in the blue)
 - Thin film, flexible form factors allowing for their use in mobile applications
 - Very attractive lighting colors and luminance characteristics
 - But...optical outcoupling losses and long lived PHOLEDs and TADF devices are remaining challenges to be solved.
- Organic lasers are excited primarily via optical pumping with features of:
 - Wavelength agility
 - Extraordinary temperature stability of its threshold and spectral properties
 - But...electrical pumped lasing remains a challenge due to large triplet and contact losses.

